

For analysis the 2.13g crude lactone acid was dissolved in 80cc hot acetone & the filtrate concentrated to about 30cc and then add acetone ether in small amount which deposited 0.35g, crop I, tan xyle. Decant from these and then distill off a portion of the mixed solvent when got a separation of 1.37g of ^{crop II} ap. lactone acid in form of heavy prisms melting at 165°

0.1904g dried in vacuum gave 0.2852g CO₂ and 0.0797g H₂O

Theory for C₈H₈O₆

C 40.90
H 4.53

Found
40.87
4.65

The filtrate from crop II gave 0.34g more pure lactone acid.

Now took 1.87g of ap. lactone acid & 1.2g powder. Noel & see Essig. same analysis & heated 9½ hrs from 110°-120° and got ^{when distilled off in vac} (other ex tr) 2.25g gum containing xyle (last other gave nothing more) then heating the 2.25g extract in Bunsen in the solution & finally crop on Bunsen in dist. to small volume got a separation of 0.29g of 2-oxymucic lactone in yellowish plates mp 228-230°

The ag. filtrate crop on Bunsen 1.45g gum which again heated with 1.0g powder Noel & see the ac. filtrate crop on Bunsen 2.6g gum + Noel (90° 30 min) from which got by other etc 0.15g more xx C₈H₄O₄ there was probably still in there some formed by 2nd test near with Noel & Essig. analysis

Proof that body C₈H₄O₄ is identical with one obtained from galactometasaccharic acid (see No 55) is as follows. Took 0.29g, crop I, solid solution & heated it.

(a) 0.2823g sub and 27.9872g H₂O (p = 1.0) and found d in 2dc tube was ± 0.0 absolutely.

(b) The H₂O solution of the 0.2823g-2-oxymucic lactone took in cold (phenolphthalein) 20.5 cc $\frac{N}{10}$ NaOH to get permanent pink coloration (theory for C₈O₄H₂-H = 20.16 cc $\frac{N}{10}$ NaOH)

now add excess in all 43.02 cc $\frac{N}{10}$ NaOH & titrate back with $\frac{N}{10}$ HCl etc got as final result a total of 39.12 cc $\frac{N}{10}$ NaOH used (theory for C₈O₄H₂-H₂ = 40.32 cc). The ag. solution of the Na salt distilled off in vac got pink again. had to add 1½ cc $\frac{N}{10}$ HCl to get rid of color now

dissolved yellowish gummy Na salt in H₂O as to make total vol = 40.60g H₂O (p = about 1.0) & found d in 2dc tube was about ± 0.0. Now try to get back 2-oxymucic lactone from ag. solution by adding 36.0 cc $\frac{N}{10}$ HCl etc got back (other ex tr) about 0.28g yellowish (?) keto by down mucic gum

(c) can't get back 2-oxymucic lactone: the yellow gum 0.28g gum in H₂O solution with ph. hydrogen acetate a reddish semi solid oily ppt soluble in Na₂CO₃ & 3% propylene glycol

H₂ did acid. may be a hydrogenate of the 2 keto acid. June 4 09 Oct 13 09, many 5 + 14-10

(55) Conversion of galactometasaccharic acid into the 2-oxymucic lactone

acid of Kiliari - det of its space formula & conversion into 2-oxymucic lactone

Took 3.35g of galactometasaccharic + 13.4g H₂O (1.21) & heated 60° for 48 hrs & got 100° warm 3.6g residue and oxide xyle. This dissolved in 80cc H₂O & add 1.0g powder CaO & heat Buns 3 hrs gave 0.77g more CaO & 0.4½ H₂O acid filtrate then digested with 0.2g more CaO till penn alk. reaction for standing over night ag. filtrate (vol 12) gave 2.4g

If Bowen & Co. order see Nov 10 29 not put in paper

i special formula of α -galactose
 The aq. filtrate from crop #4 Ca salt above given in prep. α -galactose
 which was probably a mixture of unchanged α -galactose & galactose
 2nd Oxid. expt. 6.7g α -galactose metasaccharic acid from various sources was heated
 40 hrs 40-45° with 26.8g HNO_3 1.21 + then poured into H₂O make up 1.0 L & got in leaving 10 cc
 1416 cc $\frac{2}{10}$ NaOH needed; treat rest of aq. sol with 5.3 g powder CaH_2O_7 4 H₂O (in 2 crops)
 CaCl₂ 0.4 $\frac{1}{2}$ + 5.3 g xx needles of α -galactose metasaccharic acid $\text{C}_6\text{H}_{10}\text{O}_7$ 4 H₂O (in 2 crops)
 More with 2.1g Oxalic acid 2 H₂O etc given on expt 3X with 600 cc ether only 1.0g α -galactose meta
 saccharic acid as before strong plates up to 140-150°; the rest of the acid 2.0g was
 extracted as above by same H₂O etc & given with 1.2g powder NaCl + 8 cc acetic acid 10 hrs 100-110-120°
 when distill off vacuo 90° 15 mm + take up in ether etc a total of 1.4g gum + xx lls which
 with cold ether gave 0.5g xx lls which xx from H₂O gave 0.42g cp. 2 Oxy succinic lactone
 in yellowish plates up to 230°.
 The 40g crude acid (ether extract) given once 100-120° with 0.6g powder NaCl + 4 cc Essig saure by dist
 got distill off 90° 15 mm when extd with ether etc 0.90g crude gum + xx lls which heated with
 H₂O on Buns. finally conc. in crop dist gave 0.2g more 2-Oxy succinic lactone up to 230°; the
 aq. filtrate gave 0.6g gum by acid only. (i. have 2 reaction products always) the gum is easily
 sol in cold H₂O + the xx lls are very difficultly soluble; thus the 0.62g xx lls obtained from
 all (3g) α -galactose metasaccharic acid were xx from 12 cc hot water for analysis and got
 very quickly while still hot a rapid separation of transparent very heavy yellowish plates
 up to 228°-230° (used for analysis & optical investigation). aq. filtrate crop left 0.13g gum + xx lls
 found equally pure. Finally the combined gummy acid obtained from 3g acid $\text{C}_6\text{H}_{10}\text{O}_7$ was
 found to be 0.9g & this heated again with 0.6g NaCl + 3 cc acetic anhydride 8 hrs 100-120° etc
 gave 90° 25 mm etc 1.61g gum + NaCl, from which could get only 0.680g more $\text{C}_6\text{H}_{10}\text{O}_7$ i. the gum
 is an independent & main reaction product in all case when the meta saccharic
 acids of the galactose as well as dextrose series when heated with NaCl + acetic anhydride & E.D.

0.1848g C₆H₄O₄ dual H₂O₂ in vacuo m.p. 228°-230° [α]_D²⁰ = +0.0 g/cc
0.3484g O₂ and 0.0503g H₂O
Nerv for C₆H₄O₄ Found
C 51.43 51.42
H 2.86 3.02

On mixing a rem. amt of this body with a comp. amt. of C₆H₄O₄ obt from dextro metasaccharin
(see No 54, 0.29g lot) found m.p. was 228°-230° i. bodies are identical.
Optical investigation: 0.2112g substance and 20.8874g H₂O (i. p. = 1.0+) used found d in
2 dc tube was ± 0.0 . (H₂O titration); m.p. titrated aq. sol. of 0.2112g C₆H₄O₄ using ph. tholin d
found took 15.5 cc $\frac{n}{w}$ NaOH in cold to get pink color. Theny 15.8 cc for C₅H₃O₃-CO₂H; now add in
all 38.75 cc $\frac{n}{w}$ NaOH heat then titrate back with 9.78 cc $\frac{n}{w}$ HCl etc to get as final result
28.97 cc $\frac{n}{w}$ NaOH needed (theny 30.16 cc). Now distill off H₂O in vacuo found got pink color:
had to add 1.5 cc more $\frac{n}{w}$ HCl then got light pink color to disappear then distill off rest
of H₂O & dissolving giving yellowish Na salt in H₂O & shaking up to total weight 30.15g
found d in 2 dc tube to ± 0.0 i. optically inactive C₆H₂Na₂O₄ salt D.E.D.
Now set free acid by theny 27.5 cc $\frac{n}{w}$ HCl etc to get 0.2g yellowish green easily soluble
in cold H₂O: C₆H₄O₄ can not be removed after conversion into di Na salt but got
a yellowish acid? 2 keto hydroxymuramic acid which in H₂O solution gave with ph. by hydrogen
acetate an oily yellowish ppt. sol in soda & ppt by dil acetic acid as was the case (see
with a comp. preparation of C₆H₄O₄ from dextro metasaccharinic acid No 54)
(56) Conversion of β Galacto metasaccharin (Kilian's Pan saccharin) into June 4 07
 β Galacto metasaccharinic lactone acid & det of its space formula June 11-10

Took 3.15g β Galacto metasaccharin gum 37.34g HNO₃ (1.21) heated at 60° for 48 hrs & got 95%
3.3g residue and xx of oxalic acid (of Ber 37, 3613 Kilian): this dissolved in 200 cc H₂O and 1g Borax
Calc 3th Borax gave 1.0g mod. CaO & H₂O then making filtrate with 0.15g more CaO till perm. alkali
sol + let filtrate stand for no xx of a Ca salt: trial K₂Cr₂O₇ jump when Ca salt finally came out as a gum
(vol about 100 cc) (ind. dry) heavy Ca salt Ca C₆H₄O₇ 5 H₂O + conc aq. filtr got 0.15g more: total
amount 1.67g of xx heavy Ca salt Ca C₆H₄O₇ 5 H₂O (in which Ca salt did not entirely dissolve)
1.82g: this with 0.71g oxalic acid + 200 cc H₂O in Borax (theny 0.995g) m.p. 130°-153°
& dist off aq. filter in vacuo gave 0.93g crude xx lactone acid C₆H₄O₆ (theny 0.995g) m.p. 130°-153°
which gave [α]_D²⁰ as -90° 45' found d in 2 dc tube = -7° 32' (p = 4.006): now took all of oxide acid & lactone
aq. solution (lithium): it took 55 cc $\frac{n}{w}$ NaOH in cold to get neutral (theny 54.3 cc) then adding more alkali
turning solution got acid again finally named metasaccharinic when theny (2 mols) NaOH = 108.6 cc
 $\frac{n}{w}$ NaOH had been added. The aq. sol. dist off more & take up Na₂ salt 1.292g by theny in H₂O
so as to make total weight = 32.3001g (i. p. = 4.0) & found d in 2 dc tube = -1° 41' (i. p. = -17° 48'.
hence space formula of β Galacto metasaccharinic acid is $\text{COOH} \begin{array}{c} \text{H} \text{ H} \text{ H} \text{ H} \\ | \quad | \quad | \quad | \\ \text{CH} \quad \text{CH} \quad \text{CH} \quad \text{CH} \end{array} \text{COOH}$ D.E.D.
Now add 110 cc $\frac{n}{w}$ HCl to Na salt & distill off 50° to remove & take up in 100 cc water then got back
0.98g crude lactone acid, of which 0.48g was obtained (m.p. 159 to 160°)
from this amount in conc & cooling (in H₂O).

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and the less pure body To get rid of a trace of NaCl in pure lactone, dissolved it in acetone filter; distill off then on adding sm. amt of ethyl acetate for 0.22 g more esp. lactone and m.p. 159-160°: total amt. = 0.67 g pure body $C_6H_8O_6$ which now give following constants: 0.6567 g substance and 15.7389 g H₂O ($\therefore p = 4.006$) and found in 1 cc tube = -3.97 $[\alpha]_D^{20} = -98.05$

Now took eq. sol. of 0.6567 g substance added in all 74.62 cc $\frac{20}{10}$ NaOH = theory + must have 0.8875 g $C_6H_8O_7Na$ & then distill off in vcu & make up c H₂O to total weight 22.2447 g ($\therefore p =$) & found in 2 cc tube = -1.47 $[\alpha]_D^{20} = -18.23$. Gr. back with 74.62 cc $\frac{20}{10}$ HCl etc. (peltone ex tr.) 0.6 g lactone and which 3 cc conc. only & m.p. this gave 0.97 g acetyl derivative i.e. no $C_6H_8O_4$ is obtained unless NaCl is also present -

(57) Analysis of α -dextro metasaccharine; m.p. 92° xx from ethylacetate (Dec 30° 1908) $[\alpha]_D^{20} = +6.2$ or $+8.2$

0.2113 g substance gave	0.3420 g Bar and	0.1202 g H ₂ O	Found
theory for	$C_6H_{10}O_5$		44.16
C	44.44		6.36
H	6.17		

(58) The conversion of α -galacto metasaccharine to β body by shifting of α OH in α position

Nov 15-09 of Dec 6-09 (cf N46 & 20). Jan 14-10 - Feb 7, 10.

Took the 8.2 g cp. α -galacto metasaccharine (Oct 2 + Nov 9, 09) No 51 sealed in oil bath with inside temp ranging from 190°-197° & volume temp 192-200° for 7 hrs. Got a brown colored gum & small amt of H₂O noticed at top of d.f. On adding dissolving gum in 5 cc abs. alcohol got 4.1 g α -galacto metasacch XXlle (will wash with cold abs. alc.) & then 0.25 g more. The alc. filtr. finally gave 3.53 g (100° 25 mm) gum which was acetylated with 11 cc conc. only 100-110° 11 hrs & got 5.33 g gum of which 2.55 g tar (by diff) did not dissolve in ether & got from other sol. acetylated gums on 1x hydr with H₂O & 1x hydr with 5.0 g BuOH etc 1.95 g gum which xx from 4 cc abs. alc. gum 0.63 g more α -galacto metasaccharine & the 1.15 g gum of same from 1.95 g gum which by 2.8 g BuOH etc gum 4.1 g xx salts which xx from 100-110° 9 hrs with 28 cc anhydrous gum 13.9 g acetylated gums of which 4.5 g tar remained insol. in hot ether. For 9.75 g other sol. acetylated gums which by dr. 1x c H₂O & 1x with 10 g BuOH etc 25 cc more (this BuOH) gave 5.4 g gums; on diss. these in 100 cc abs. alcohol got 1.67 g xx α -galacto metasaccharine & then on dist off overstanding in small and galactol for 0.38 g more of similar xxlle (2.05 g total), alc. filtr. now gave back 3.3 g gum which with 8.0 g BuOH etc gum 12.3 g salts & these xx from 50 cc alcohol 98% gave rapidly 5.02 g β -galacto metasaccharine BuOH xxlle in desben & Schwenm Kneten m.p. 130-135° - The alc. filtr. gum was more xxlle: set free gum by 5 g BuOH & got 3.95 g BuOH & 1.9 g gum which converted by dryness into salty gum no xxlle: set free gum & acetylate 7 hrs w/ with 6 cc anhydrous got 2.68 g acetylated gums - & found this left 0.53 g more tar on distillation with ether. The hydr. of the sol. acetylated gums 1x c H₂O & 1x with 5 g BuOH etc

(80) ($b = 3.84$),
gum back 1.2 g gum whose $[\alpha]_D^{20} = -14.4$ only, 'in another body present makes the
& β metacaccharins whose $[\alpha]_D^{20}$ must be either slightly - or +.

The 1.2 g gum dissolved in alcohol m. and gave 0.2 g α galactometacacch. x x lte & the
rest of gum left in ale filtr 0.95 g (act. det.) gum with 2.3 g Bu cin de 3.2, salts whose x x
from 6 cc alcohol gave 0.83 g non β galactometacacch. Bu cin x x lte up to $120^\circ - 137^\circ$
the ale filtr was now lost by an accident: unknown body, $\text{CH} = \text{CH}$ in it, $\text{CH} = \text{CH}$ in it,

could not be further studied.
Proved that β galactometacaccharin was formed; took the 2 crops of crude Bu cin salt
5.02 g and 0.83 g $\therefore 5.85$ g and got by xx from 2 cc H₂O + 35 cc abs. alcohol 4.09 g

destr x x lte up to $130 - 137^\circ$ $[\alpha]_D^{20} = -26.2$ i.e. 1.018 g salt and 24.445 g H₂O ($p = 4.0$)
and found α in 1 de tube = -1.06 . (alc. filtr discarded cont 16.5 g salts)

The 4.09 g β salt gum with 2.0 g Bu cin etc 2.58 g Bu cin and 1.16 g gum (alc ext) which
could not be got solid: digest it Bu cin with excess strychnine 8 1/2 hrs Bu cin found 1.88 g dissolved
then got 3.75 g gum salt which xx from 1 cc H₂O & 8.5 cc alcohol gave 1.63 g and 0.46 g crop II

op. β galactometacaccharinic Strychnine in: flat needles up to $120^\circ - 130^\circ$.
Crop I gum $[\alpha]_D^{20} = -23.43$ i.e. 0.9903 g salt and 23.698 g H₂O $p = 4.013$ found α in 2 de
tube = -1.9 [cf. No 59]; the 2nd crop 0.46 g β salt was obt by digesting ale filtr crop I

again for 9 hrs Bu cin with excess prod strychnine etc. (alc filtr from crop II salt was discarded).
Now take the combined crop I & II of β Strychnine salt got by 1.0 g Bu cin etc 1.24 g
strychnine & 0.68 g (alc ext) β gum which gave $[\alpha]_D^{20} p = 3.995 = -46.9$ i.e. β salt
found α in 2 de tube = -3.77 . Now distill off H₂O got back 0.62 g gum & this gum with

0.7 cc ph. hydrazine & 0.7 cc alcohol within 1 1/2 hrs a mass of xx hydrazide; after standing at $85 - 90^\circ$
and alk acetate till all powder got 0.51 g yellowish β metacaccharine ph. hydrazide up to $85 - 90^\circ$
(analyzed for C, H & N by Upson). This gum $[\alpha]_D^{20} = -2.23$ i.e. 0.5017 g salt and 12.0702 g H₂O p
= 3.991 found α in 1 de tube = -0.09 . The hydrazide decomposes in all cases rapidly on

standing [cf. No 59] Q.E.D. identity with β galactometacaccharin from d. galactose 33%.
(59) Galactose + 33% NaOH + final, continued { Jan 14-10 Feb 7-10 and May 19-10
see 18-09

the finding the 2.7 g gum ($[\alpha]_D^{20} = +4.7$) with 6.5 g Bu cin got 9.6 g salts which xx from 1 cc alcohol
got solid with x x lte crop I up to $100 - 135^\circ = 2.82$ g probably a mixture of α & β galactometacacch salts.
to prove this took the 2.82 g crop & got Bu cin etc 0.83 g gum (alc) with $[\alpha]_D^{20} = +2.72$ i.e. this can't
be such a mixture but body was probably β isosaccharin. Now got back gum 0.33 g

with acetic ether got 0.52 g gum $[\alpha]_D^{20} = +3.95$ which with 1.2 g Bu cin etc gave β isosaccharin should give a
crop I salt up to $140^\circ - 150^\circ$: previous salt (discarded) N.B. the β isosaccharin should give a
previous salt like l. anthro is gum & up to should be over 180° - but too small an amount
present to isolate it as a pure product. The ale filtrate from crop I salt gave back
with Bu cin etc 4.02 g Bu cin & a gum which ext R.F. with 1 de ether gave
1.48 gum $[\alpha]_D^{20} = +7.79$ ($p = 4.0$) and in ext

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($\therefore$ discard $p = 4.0$),

residue still further with ethyl acetate got $\frac{0.37g}{7\text{ths}}$ more gum from $\neq p = 4.0$.
The $1.4g$ gum $[\alpha]_D^{25} = +9.99$ boiled R.F. H₂O with $2g$ Ch₂ in 20cc gave $4.4g$ gummy salts
which dissolved easily in 10cc abs^l alc & gave nothing on evoling $\therefore$ α & β isosaccharin can't
be present i.e. has been entirely removed in previous operations. Got back But₂ the
etc $0.87g$ gum (see also extr) & treated with $2.0g$ Bu₂en + xx salt obtained & got
 $0.37g$ gums salt crop I m.p. $190-195^\circ$, which must have been the β isosaccharin
discard material at this point
Bu₂en (see p 80).

The $4.85g \pm$ gum (bottom p 69) heated R.F. with 30cc ethyl acetate shake & evl gave
 $2.37g$ soluble gum $[\alpha]_D^{25} = +2.347$ $p = 4.0$ & $0.87g$ acc. det
and $\frac{2.80g}{2.80g}$ insol gum $[\alpha]_D^{25} = -1.173$ $\rightarrow$ treated for α factor (discarded).
On extr the $2.37g$ sol R.F. 2x with 10cc ether from all but $1.0g$ (by diff) dissolved & found in
ether solution $1.5g$ gum with $[\alpha]_D^{25} = +5.52$; the $0.87g$ sol gave $[\alpha]_D^{25} = -2.72$ $[\alpha]_D^{25}$.
This degraded with $3.6g$ Bu₂en gave $5.18g$ salts from which got on xx from 6cc abs^l alcohol
 $0.7g$ xx needles m.p. $188-205^\circ$? β isosaccharinic Bu₂en probably: got crop II $0.82g$
m.p. $155-165^\circ$. discarded all material at this point - but observations clearly show
presence of β isosaccharin as a + st. saccharin with high melting Bu₂en salt (as this
body is & oxy methyl l. extr from $6g$ gum) besides & β Galactometasaccharin.
 β Galactometasaccharinic Bu₂en: had a total of $51.89g$ and β Bu₂en salts as follows (see 11051)
 $18.28g + 7.24g + 11.75g + 4.35g + 10.27g$. On xx this from 17cc H₂O + 20cc abs^l alc got $27.16g$
crop I β Galactometasacch Bu₂en m.p. depts $130-140^\circ$ $[\alpha]_D^{25} = -25.6$ ($p = 4.015$): this gave
with $12g$ Bu₂en $17.77g$ Bu₂en + $7.47g$ and β Galactometasaccharin gum (also extr).
it was now degraded R.F. $14\frac{1}{2}\text{hrs}$ (in 2 operations providing 2nd time) with an excess ethyl wine & found
 $13.16g$ Strophurine had dissolved & got $26.25g$ xx salts which dissolved in 5cc H₂O + 4cc alc
gave $11.42g$ sp. β Galactometasacch. Strophurine xx the m.p. depts $125-130^\circ$ $[\alpha]_D^{25} =$
 -23.48 $0.9136g$ salt and $21.9049g$ H₂O: $p = 4.003$ for 2 in 2dc tube = -1.90 .
got $30.9g$ crop II m.p. $115-125^\circ$ pale filtrate boiled again with Strophurine & got $1.27g$ and then
gave $3.64g$ more crop III β Strophurine salts = $17.15g$ with $9.0g$ Bu₂en etc $11.0g$ Strophurine
got by treating crops I II & III β Galactometasaccharin gum which did not get solid on seeding with
and $5.46g$ (also extr) β Galactometasaccharin gum which did not get solid on seeding with
(β xx obtained Nov 25 08) but gave $[\alpha]_D^{25} = -49.7$ $p = 4.0$ (found 2 in 2dc tube = -4.02).
" gum is a mixture of lactone + salt? free acid or other anhydrides. Now make up with H₂O to
50cc & mix with 10cc for ph. test & expt and 60cc for rotation of the salt.
Rotation to salt: 60cc $\therefore 0.6553g$ lactone should give $0.8170g$ $C_6H_{11}O_6Na$ $\therefore$ treated with 40.45cc
 $\frac{1}{10}$ NaOH & heat on Bu₂en. then should get pure neutral (to be used)
solution at reduced pressure & dissolve in H₂O

to total weight 20.425g and found ($\therefore p = 4.0$) α in 2 de tube = $+0.31$ hence $[\alpha]_D^{25} = -3.833$.
Phenylhydrazine 100 cc of solution $\therefore 109$ g distilled off 100° 20 min & local gum in cold with ice ph. hyd
+ ice alcohol soon got entirely solid with snowwhite nodules (under scratching). after 20 hrs slowly at
all ethyl acetate break up lumps got 1.61g almost white ph. hyd reagent mpt $85-90^\circ$
(analyzed for CH+N by Upson) - which is a very unstable by product & can not with out much loss
be xx from (ale or alcohol & acetic ether. $\therefore$ det $[\alpha]_D^{25}$ of crude product directly after drying
H₂O in vacuo as - 1.92 is 0.5898 g sub and 14.190 g H₂O $\therefore p = 3.99$ & found
 α in de tube = -0.155 . Rest of β galactometasacch. gum 3.76 g was used up in
a 2nd α id to β galactometasaccharine acid May 19-10 method book II we worked up
or described in these sheets. (p 81).
The de. filtrate from the 27.16 g crop I β galactometasacch Bmeim still contained 24.73
g Bmeim salts & gum on xx from 4 cc H₂O & 30 cc alcohol 9.85 g crop II salts mpt $110-140^\circ$ is a mixture of
 α & β galactometasacch. Bmeim $\therefore$ treat all with 10 g Bmeim got 16.11 g Bmeim & 6.55 g gum (slightly
which dissolved in 7 cc ale & 20 cc then 6 cc ethyl acetate gum 0.87 g α galactometasaccharine xx the
mpt $135-140^\circ$. Ale. filtrate gum back 5.4 g gum $[\alpha]_D^{25} = -11.88$ ($p = 4.0$) which with 13.1 g
Bmeim ale gum 20.1 g gum salts from which got 7.83 g crop I & a pure Bmeim salt mpt
 $135-150^\circ$ on rexx this from 2 cc H₂O & 35 cc alcohol got only 3.88 g β galactometasacch. Bmeim
mpt $135-140^\circ$ (keep); the ale. filtrate gum 3.8 g xx salts which xx from 15 cc absol. alcohol gum
 17.8 g of a Bmeim salt and xx H₂O safely drying H₂O in vacuo (wt 1.7 g) melted from $150-153^\circ$
and had $[\alpha]_D^{25} = -21.72$ (this gum Bmeim de 0.468 gum $[\alpha]_D^{25} = -50.69$ ($p = 4.0$) situation of the gum
Nee ale. filtrate from crop I 7.83 g salt should contain about 11.5 g salts for each xx from 30 cc alcohol
gum 3.57 g crop II salts mpt $110-120^\circ$ $[\alpha]_D^{25} = -19.75$ is mixture of α & β metasacch Bmeim
now discard all material as could go no further.

It is obvious that above was a mixture of α & β metasaccharine & of α & β Metasaccharins
(60) On the conversion of α dextrometasaccharine to β by shifting { May 25-09
Apr 3-10
Apr 10-10 }
8.0
Took 6.18 g xx & lactone from 1.18 g op Ca salt (V₆ 52) heated d.f. in a solder bath bulande lamp
 215° ten side lamp $200-210^\circ$ for 9 hrs: a little H₂O noticed in d.f. Now acetylate with 20 cc acetyl anhydride for 9 hrs
w₂ got 9.73 g acetylated gum of which 3.0 g (by diff.) in hot ether. got 7.07 g etherol acetylal
gum which by de. 1 cc H₂O $\rightarrow 4.72$ g when with 10 g Bmeim + 35 cc H₂O gum 3.99 g (ale ext) gums, which
digested $9\frac{1}{2}$ hrs with excess Phychrome took up 6.48 g & gum 11.25 g Phychrome salts mpt $180-190^\circ$; ale filtr
+ 30 cc ale dist gum 2.67 g and α dextrometasacch. Phychrome mpt $180-190^\circ$; ale filtr
distilled off & heated again with 1 cc H₂O & 20 cc alcohol after 1.35 g crop II crude α & dextrometasacch
gum 9.15 g salts which xx from 170-180. Crops I & II = 4.02 g, gum with 20 g Bmeim 1.28 g
Phychrome crop II mpt $170-180^\circ$ (ale ext) from which got by CaH_2 ale de a total of
and α dextrometasacch. (pale ex l_r)
1.05 g op. anhydrous xx Ca salt $\rightarrow$ D.F.D.
The ale filtrate from crop II Phychrome salt gum with Bmeim etc 3.97 g Phychrome and
 2.43 g crude β dextrometasacch. gum which on searling got a fair amount
by xx the but and the sublt traces of α present as shown
by what follows:

Now acetylate with 20 cc acetyl anhydride for 9 hrs
got 7.07 g etherol acetylal
gums, which
mpt 180-190; ale filtr
crude alpha & dextrometasacch
from which got by CaH2 ale de a total of
Phychrome salt gum with Bmeim etc 3.97 g Phychrome and
2.43 g crude beta dextrometasacch. gum which on searling got a fair amount
by xx the but and the sublt traces of alpha present as shown
by what follows:

Treated gum with 5.91 g Buene + got 9.1 xx salts which xx from rose alcohol soon got solid with
xx lls on seeding with 5 dext to Buene xx lls: got 4.4 wpt dext 120-150° after drying in vacuo
with $[\alpha]_D^{20} = -22.03$ i.e. 0.8608 g salts to 20.6842 g H₂O ($p = 3.995$) and found α in 2 dc tube = -107°.
On setting free gum from this by wpt Buene etc got 2.74 g Buene and 1.32 g gum (also xx lls) which
would not get entirely solid even on seeding with 5 dext to Buene xx lls: converted it into
Styrene salt Buene 8 hrs took up 2.2 g Styrene gave 3.63 g gummy salts from which by
dissolving in 1 cc H₂O twice alcohol seeding with salt could get no separation of xx salt: α found in 1 dc tube = -0.12
free gum by Buene etc took 1.15 g gum where $[\alpha]_D^{20}$ was -2.97 ($p = 4.0$) α found in 1 dc tube = -0.12
∴ another substance which must be strongly basic, probably. $CH_2=CH-CH=CH_2$ must be
present. Now label of H₂O in vials + find gum with 1 cc alk + 1 cc ph. by drying + got 3 days

O.T. seeding with 5 ph hydr. a total
which was obviously identical with 3 dext to Buene xx lls in phenylhydrazine (see No. 1)
for it xx from alcohol in transparent plates.
The alk filtrate from crop I Buene salt gave 3.85 g salts from which by xx from 10 cc alcohol
got on long standing 1.83 g more Buene salt $[\alpha]_D^{20} = -23.3$ wpt 130-140°
this by Buene etc gave 0.47 g gum (also xx lls) from which by 5 sec ph hydr + 0.5 cc alk got only 0.01 g
ph. hydr. wpt wpt -100° as above (∴ write + keep).
It is obvious from all this that best way to grow α + 3 dext to Buene xx lls is to
space corners differing only in position of OH group in α carbon alone is to
convert the 3 bds by heat 200-220 into the α as it can be isolated from as shown above.
(61) Oxid. of d. 1,3 dioxystyrene acid Nov 22, Dec 16 09 Jan 12-19
TAK the 5.85 g ep. lactone gum
5.75 g H₂O3 spgr. 1.21.

(April 14-09) (No. 34 p 46 top) $[\alpha]_D^{20} = -40.37$ + dissolved it in 23.2 g H₂O3 spgr. 1.21.
Oct 2-09 (No 50 p. 66 bottom)
+ heated 48 hrs in bulk kept between 60-64° then distill off in vacuo finally at 90-95° at 25 mm Hg.
got 4.1 g residue cont oxalic acid xx lls. Dissolved this in H₂O + stirred Buene with 1.0 g powder Cal got 2 wpt
1.6 g Cal wpt H₂O and some Ca salts. The filtrate heated with 0.1 g more Cal powder finally the powder was
solution upon cone. to see if can get a xx Ca salt. n.g. ∴ had to ppt Ca salts from conc aq. sol by
alcohol and 4.7 g ppted Ca salts washed with alk + ether and dry which lost 0.45 g in weight and H₂O
in vacuo 2 days then gave 21.36% Cal on equation (i.e. 0.2327 g gum 0.0497 g Cal 2x equation) ∴ lost
rest of salts 4.038 g with H₂O to total weight 20.95 g ($p = 4.0$) found α in 1 dc tube = +0.28 $[\alpha]_D^{20} = +6.925$;
now ppt Ca by 1.941 g oxalic 2 H₂O + got (also xx lls) after doubling off H₂O at 25 mm Hg 2.85 g gum
which was not digested Buene. 5 hrs with 124 g Chlorine (leaving 2.28 g wool Chlorine). The
alk. solution in conc gum in 3 crops a total of 6.37 g diff sol. in H₂O Chlorine salts wpt 160-50
(5.3 g + 0.82 + 0.25 g). The salts left in H₂O (specially soluble from conc to small volume) were processed by
xx from alcohol (30 cc, after doubling off H₂O in vacuo 95° to leave some unchanged d. three salts
as got 0.75 g salt wpt 171-2° } got from rest by Buene etc 1.05 g gum containing
0.625 g wpt 174° }
xx lls ~~which~~ of malonic acid
(see next page)

The 6.37g diff sol in H₂O Chlorine salts gave with 5.0g BuN₃ 5.2g Chlorine and 1.15g gum containing XXll. This was extracted R.F. 2x with wce ether when all except 0.15g by diff had dissolved. ^{in one} The other solution in WB to a small volume got a separation of 0.38g ap. 13 Dioxyl glutaric acid in brown white heavy XXll melting sharply at 135° to a liquid and giving off water vapor. The other filtrate gave back 0.6g gum which was probably a mixture of d. malic and 1,3 Dioxyl glutaric acids. ~~These~~ but these could not be separated & discarded. The 1.05g gum obtained from easily sol. Chlorine salts dissolved easily except 0.11g tar in hot ether (3x wce) gave 0.45g gum containing a xx subst melting in xx at sides of d.f. which was probably malonic acid - from digest from BuN₃ with 0.5g powdered stone ag. filtr (Vol 30cc) to a smaller volume got a rapid separation of 0.4g diff. sol. Ca salt ~~lost material on heating it with 0.25g oxalate~~ OK. -

Wgt xx of malonic at this point. The 0.38g ap. 13 Dioxyl glutaric gave $[\alpha]_D^{20} = -2.614$ i.e. 0.3782g and 9.1404g H₂O (i.e. $p = 3.974$) and found $[\alpha]_D^{20}$ in 1 cc tube = -0.105. Titration like a free dibasic acid: it took up at once in cold 45cc $\frac{n}{10}$ NaOH solution remained acid (litmus): now ran in a total of 49.3cc $\frac{n}{10}$ NaOH solution remained alkaline on long boiling in WB. Theory for C₅H₈O₆ 0.3782g = 46.12cc (and for same amount C₅H₈O₅ would be 51.81cc). The alkaline solution was now distilled off in vacuo and the xx residue left theory = 0.4798 C₅H₈O₆Na₂ with H₂O to total weight 11.990g acid found & in 1 cc tube was +0.9 ∴ $[\alpha]_D^{20} = +22.25$. To get back free 1,3 Dioxyl glutaric acid again the residue was now treated in ag. sol. with 50.3cc $\frac{n}{10}$ NaOH (i.e. 1cc $\frac{n}{10}$ NaOH in excess) from distilling off H₂O in vacuo finally heating to 95° at 25 mm etc got back by ether etc 0.38g gummy acid only ∴ a change took place probably due to presence of excess NaOH at 100° [cf. gelatin & dioxyl]. This gum broke with Chlorine took up 14g alkaline solution, ag. filtr & got a separation in all of 153g xx stuff sol in H₂O salts (and last ag. filtr and wash nothing) ∴ the gum was now set free from these salts by 1.0g BuN₃ got back 3x wce hot ether twice. The gum gum finally with 1.35g only 0.060g 1,3 Dioxyl glutaric XXll had a 0.25g gum which was probably a mixture of 1,3 Dioxyl glutaric and of d. 1-oxyl glutaric acid. The gum gum finally with 1.35g BuN₃ etc an xx salts obtained from 5cc alcohol ^{whereas the} H₂O. The BuN₃ salt of d. 1,3 Dioxyl glutaric after drying H₂O in vacuo between 120-125° - of Kiliari paper of l. 1,3 Dioxyl glutaric

acid - Res 40.1238 - 46.25 = 1910 2nd oxidation was now necessary to get d. 1,3 Dioxyl glutaric acid for analysis. Took 4.851g lactone $[\alpha]_D^{20} = -36.8$ [June 22, 10 of No 68] + 19.4g HNO₃ 121 heated 45-50° for 48 hrs. ^{then} made up e H₂O to a clear thick powd CaH₂ tell alkaline solution (gave smelly) & got 1.73g moist Ca lactate H₂O and on and CaH₂ & alk. filtrate was then come to an all volume in WB, & speed by alc etc gave 3.75g an dry w 3.32g means dioxyl Ca salts - latter containing 22.91% CaO (0.2051g salt gave an equivalent 0.0475g CaO). ∴ most part of salt 3.1625g with 1.630g oxalate ^{was} got on heating ag. filtrate with after determining $[\alpha]_D^{20}$ of a portion of Ca salt ($p = 4.0$) to be +9°92)

(85)

The University of Chicago

8.4 g Chinine etc a total of 5.38g diff sol in H₂O. Chem salts mpt 165-7°
More pure with 3.0g Barb₂ etc 4.17g Chinin + 3x500 Ethyl R.F. on conc a total of
0.25g d. 13 Diox₂ glutaric acid mpt. 135° (that H₂O in vacuo) treated with cold ether
0.2105g substance given 0.2830g Coac and 0.0950g H₂O

Heavy for C₂H₄O₆ Found
C 36.59 36.65
H 4.88 5.01

302 0.55g given from other filtrate which with 3.25g Buacm etc gave 4.65g xx salts mpt
xx from 12 cc absol. alcohol gave 2.09g sup I salts mpt 130-160° opaque nodules, which
on digesting with 20 cc absol. alcohol left 0.63g insol. d. 1,3 Diox₂ glutaric Buacm. (?)
mpt 235-238° [Keep] - dis and rest of salts? Buacm d. 1-10x glutaric

present but no indication of d. Malic acid. Pre 16-29
(62) Properties of d. Malic acid, free acid, Na salt, Buacm + Chinin salts.

Solubility, d. Malic acid, 2g. 1. Malic acid dissolves easily R.F. in 5 cc ethyl acetate on cooling came out again in part in
xxlle; 2g. l. Malic acid dissolves easily R.F. in 100 cc Ethyl on dist off solvent in WB to a small
volume got a separation of 0.84g heavy xxlle mpt 100° (in vacuo dried) with [d] = -2.16 i.e.
took 0.822 g and 19.7058g H₂O (p = 4.004) of pure d in 2 cc tube = -0.175.
Titration on adding 122.77 cc of NaOH solution became very slightly alkaline (Thom = 122.7 cc) on
distilling off H₂O in vacuo & taking up Na₂ salt. Thom = 1.092g to total weight 27.30g with H₂O
∴ p = 4.0 found d in 2 cc tube was -0.675 ∴ [d] of C₄H₄O₅ Na is -8.35.

Chinin salt, 1g l. Malic + 5g Chinin Barb₂ gave on letting solution (Vol 200 cc) stand
a separation of 4.50g sup I, l. Malic Chinin in transparent xx needles mpt 182 (Keep)
∴ 67% comes out + salt is fine for distilling mass of d & l Malic (see No 63),
The salt on xx well 1.61g from a hot sol. of 1 cc H₂O & 10 cc alcohol coming out in transp
needles in very large amounts.

Buacm salt, 1g l. Malic + 5.88g Buacm on Barb₂ with H₂O quickly gave a clear solution
+ undist of H₂O in vacuo found the 10g salt dissolved in 10 cc Hot H₂O & on cooling got a separation
of 1.67g xxlle in transp. needles mpt 220° (Keep) ∴ only 50% comes out from H₂O -
The salt however is very very diff sol. in boiling alcohol R.F. & on cooling got a separation of
heavy transparent needles -

Ca salt, took 1g l. Malic acid + 0.5g pure Ca on Barb₂ & got rapidly a (clear solution Vol 175 cc
+ on filtration some cooling got no sep. of xxlle; but none on Barb₂ only dries the Ca salt
separates out rapidly in form of heavy xx granules; got (in 3 crops) a total of 1.17g (Thom
1.28g) but must find way to dry mass other take up in cold water to get all of Ca salt
out as it stays in eq. solution to large extent other wise.

(63)

(86)

April 18-10

OK H
H H H H H

The absolute space formula of d. 1,3 dioxyl butyric acid is $C_4H_6O_4$

Took 50g ^{xx} 1,3 dioxyl butyric acid with 1280 1320 from various sources
 also from previous l. and from l. xylene expts & boiled with 200cc H₂O & 80g BaCl₂ then BaCl₂ & got 2.4g gum (cold ether ext.). This boiled with H₂O & 80g BaCl₂ in (thru 86.5g) gave an acid solution & then added
 2.4g gum (cold ether ext.). This boiled with H₂O & 80g BaCl₂ in (thru 86.5g) gave an acid solution & then added
 off so as to leave 100cc H₂O & 102g salt for 34.34g exp Ixx plate of d. dioxyl butyric BaCl₂ (dioxyl
 H₂O & in various all xx H₂O off). On xx this from 44cc H₂O got 26.48g xx plates according which
 lost 3.23g H₂O on drying H₂O & in view to constant weight 12.20% (theory for 4H₂O - 12.30%)
 the anhydrous salt gave $\Delta T_p = 20.58$ i.e. 11720g substance + 282293g H₂O $\therefore p = 3.987$ gave ΔT_p in 20 tube = -1.66.

Now took 23.25g of the anhydrous salt with 10g BaCl₂ & 19.2g BaCl₂ & 4.17g gum. This
 oxidized with 16.8g HNO₃ 1.21 for 45 hrs at 45-50° then pour into H₂O & make up to 1L & titrate
 around much for HNO₃ present for 20cc of sol required 26.8cc $\frac{n}{10}$ NaOH & on looking at the
 eq. sol with 50g of pure CaCl₂ (this BaCl₂) got no gas used CaCl₂ on concn of. But got no
 xx of Ca acetate on adding alcohol to concn of sol. got no ppt of Ca salts i.e. pptd Ca⁺⁺ by
 8.2g oxalic 2H₂O on distilling off in vacuo finally in a small d.f. when much HNO₃
 went off (bath 70° & normal air) - got 5.3g residue and xx the oxalic & on dissolving in H₂O
 to 500cc & take 10cc found 9.86cc $\frac{n}{10}$ NaOH needed: that rest of aq. sol. with 1.8g pure CaCl₂
 on boiling w.B. some of. fell to small volume & pptd Ca salts by alcohol etc; got 5.14g mass
 dried salts (cont. 21.72% CaO by analysis) i.e. total rest 4.9535g with 2.4g oxalic 2H₂O
 & then heat aq. filtr with 12.3g Chlorine got (xx from H₂O) 4.74g Chlorine d. acetate

rept 165-170° in 200g probably containing some other salt which with 3.0g BaCl₂ &
 gave 3.9g Chlorine and 0.65g xx the (other ext. R.7). From this got on powdering & washing on up.
 with sun. and dry other 0.38g d. Malic acid rept 98-100° with $\Delta T_p = +8.07$ i.e. 0.3743g salt
 acid. 8.9213g H₂O $\therefore p = 4.027$ gave ΔT_p in 100 tube = +0.125.
Titration: Use H₂O solution of the 0.3743g acid (solution) took up in cold alone 55.9cc $\frac{n}{10}$ NaOH (= theory)
 but very faint bl. from alk. reaction & by theory 0.4971g C₄H₄O₅ Na₂ salt would be present. Result
 of H₂O in vacuo & make up c. H₂O to total weight 13.7576g (i.e. $p = 3.613$) and found ΔT_p in 100 tube
 = +0.30 when $\Delta T_p = +8.29$ (cf. l. lit. No 62 - 8.35).
 Now add 53.9cc $\frac{n}{10}$ HCl to C₄H₄O₅ Na₂ salt & got back lost off H₂O in vacuo & to R.7 with other
 0.32g d. Malic acid. Took 0.1g of this and mixed with 0.1g l. Malic (No 62) & dissolved in H₂O
 and got distilling off H₂O & 0.2g d. Malic acid rept 115-125° - or (after washing & warming pour xx)
 got 0.1g d. Malic acid rept 115-125° (keep) & E.D.

(64) Oxidation of Isosaccharin by HNO₃ etc of Kilian. Ber 18 2514 38 2672 3624
 40 1238

Tried logit mass & l. 1,3 dioxyl butyric acid from d. Isosaccharin & HNO₃ directly.
 Expt I 10g d. Isosaccharin rept 94° & 40g HNO₃ 1.21 heated w.d.f. 48 hrs at 60-65° then added 2/3
 H₂O & HNO₃ temp near 50° at 25 min got 10.31g gummy residue which boiled out at
 130° for 10 hrs with 50cc H₂O gave much Ca got back 100° & mass 7.45g gum and xx oxalic.
 Boil this gum with H₂O & 2.5g pure Ca and got 1.57g insol CaCO₃ 1/2 H₂O: concn aq. filtr to 150cc
 but for no ppt i.e. add 0.2g pure Ca heat Bx B this when made all CaO had dissolved
 concn to 50cc only got no ppt of mass dioxyl butyrate of Ca
 i.e. pptd Ca by 4.5g oxalic & got back 6.55g gum

3.9g gum with a xx sublimate. Made up to 1L with H₂O + took 25 cc (0.975g) of
found 8.8 cc $\frac{n}{w}$ NaOH needed for neutralization (5.2 cc in vol) $\therefore$ heated with 11.2g Chlorine
for a total of 7.28g diff. sol in H₂O Chlorine salts in 3 crops, which with 4.13g
obtained from yellow Ca salt makes 11.41g Chlorine salts of meso + 1,3 dioxo glutamic acids,
[cf 18g xph which gave 12.39g salts]. The 11.41g heated with 7.0g BuOH 2 cc gave 0.42g Chlorine
+ 1.71g gum (acetic ether xh + will sweep over ex). This gum heated with 0.9g CuH₂ 4 hrs BuOH
gives an alk. sol. but on cool got a gummy Ca salt only which could not be banded $\therefore$ heated
with 1.5g xph 2 hrs for back 1.6g gum which boiled with 8.65g BuOH gave 11.25g xx
salts which heated with 30 cc abs. alcohol distilled ~~distilled~~ simply on standing gave 4.10g crop I,
xxle up 120-220°. when 1.8g crop II up 240° which looks like a mixture of 2 bodies.
These crops are probably meso + 1,3 dioxo glutamic

(65) Oxidation of crude 1,3 dioxo glutamic acid { May 14 09
of 13 09
Jan 5 10

Took the 5.3g gum [May 14 09 No Turkish lead stored in alcohol at summer &
hydrolyzed to his BuOH with 30 cc H₂O, then digested K₂C₂O₄ etc for back 4.9 gum with 1d) $\frac{n}{w}$ =
-28° 01. i.e. found d in 1d tube = -198 (p = 4.167). Now divided into $\frac{2}{5}$ this & $\frac{1}{5}$ the other
latter with 0.5 cc ph hyd & 0.3 cc alcohol but got 24 hrs v.t. on adding ether etc a gum only $\therefore$ neg.
as body is obviously a mixture of dl 1,3 dioxo glutamate and may also contain d. thio 5 gum but
not likely to any appreciable amount. Took the $\frac{2}{5}$ this $\therefore$ 4.5g lengthen 5 gum and
heated with 18.0g HNO₃ (1,2,1) at 60° for 48 hrs + then pour into H₂O + make up to 1L state
30 cc + found 18.41 cc $\frac{n}{w}$ NaOH needed $\therefore$ heated rest of ag. sol with 2.7g powder Cu (2.53g = thing)
5 hrs BuOH vol 2L for 1.45g wool Ca salt as a heavy granular powder - obviously calcium mesodioxo
gave a separation of 0.93g Ca salt (1). The ag. alkaline filtrate on evap. to a small volume
- Glutarate. This was dissolved in 300 cc hot water + digested with 0.53g H₂C₂O₄ 2H₂O + got on
distilling off in vacuo + taking up residue in Acetone, heates 0.24g wool Ca salt 0.43g
xxle from which got 0.2g melting from 110-115° strong xx this at 140° $\therefore$ probably a mixture of
acid (see No 66). The 0.24g wool Ca salt (in acetone) was easily sol in cold water for heating with
0.1g gum oxalic etc gave 0.2g xxle up 110-120° strong H₂O c 140° $\therefore$ probably a mixture of
oxalic + meso 1,3 dioxo glutamic acids - ~~only~~ sol in cold ether + i.e. ~~at 140°~~ $\therefore$ probably a mixture of
considered to be meso + 1,3 dioxo glutamic acids. (N.B. dl Malic may be present in the meso Ca salt)
The ag. filtrates from 0.93g meso Ca salt gum in filtering with alc etc 4.53g air dry Ca salt
or 3.9g meso dried Ca salts containing 18.15% Ca $\therefore$ treat rest of ag. sol with 2.1g of oxalic 2H₂O
for 2.55g mobile gum (Acetone extr). This digested with Chlorine for up 8.53g + gave alcohol
of 4.74g (diff. sol in H₂O) sweeps Chlorine salt. The 4.74g salts gave with 5.0g BuOH 2 cc 3.7g Chlorine
+ 0.53g gum (Acetone extr) which with 2.9g BuOH etc gave 3.4g xx salts which warmed with 30 cc
abs. alcohol left 1.05g wool BuOH salt up 223° (? dl Malic is BuOH) $\therefore$ this an xx from 2 cc
H₂O + 2 cc alcohol gave xx prisms of dl Malic BuOH + leaving opaque residues of meso 1,3
dioxo glutamic

(66°) The oxidation of d. anthro gum to Meso 1,3 dioxysuccinic acid. May 19-20

Took 16.35g Bismuth salt mpt 198°-200° Δ 75 = -23.2 d 4.96g hydrated Bismuth salt of d. anthro gum Bismuth Δ 75 = -21.81 (May 2-10) + got c. 8.0g Bismuth salt 16.02g Bismuth salt
and 4.82g c.p. d. anthro gum (ale xtr). This gum on heating 36 hrs at 45-50° with
20g HNO₃ 1.21 + then add the smoke up to 1/2 ale. 10cc took 18.73 cc $\frac{1}{2}$ N.H. $\therefore$ heated not of
aq. sol with 3.5g powder. Cat. the 3 hrs Bismuth salt 1.94g resol Ca Cro₄ H₂O and here Ca₂ H₂SO₄
Ca salt (most boiled this out long time Bismuth with 1L H₂O & heated aq. fllr to a small volume got 0.74g (in 2 cups)
mean Ca salt possibly present. On conc. the alkaline aq. filtrate to a small volume got 0.74g (in 2 cups)
of a (xx) Ca salt which took about 1L water in Bismuth to dissolve ($\therefore$ a very diff sol. salt). The solution was
sat. C. Co. then boiled ale + filtrate then conc. in Bismuth gum on conc. to a small volume (?)
anhydrous Calcium mesodioxysuccinate 0.51g as a heavy xx salt which lost nothing on
drying over H₂O in vacuo. These were now boiled with 70cc H₂O (w/2 all in solution) then 0.290g
oxalic acid added & the mixture digested & shook long time to ensure complete reaction to Ca Cro₄
etc. Then boiled off aq. filtrate in vacuo stake up R.T. in mixed Ether (1L) got on conc
this to a small volume 0.16g + 0.02g meso 1,3 dioxysuccinic acid (and lactone acid)
in the form of a tabular often cubical mass of xx salt kept 120-130° losing H₂O & filling solid &
then not melting until 170° melting point of meso lactone acid (Kilian: 168)
0.1789g acid gum 0.2563g Co₂ & 0.0763g H₂O Found
Heavy for 36.59g for C₅H₈O₅ 39.07g
4.88g 4.74g $\therefore$ about 2g. pts free
acid & lactone acid

It took Kiliani on 20 grams to get the lactone acid:
The aq. filtrate from 0.74g meso calcium salt on filtering with ale ale gum 4.23g meso dried Ca salts
containing 2316% Ca₂ $\therefore$ had not 4.0875g with 2.130g oxalic acid got on heating aq. filtrate
with 11.0g Chlorine etc. At total of 6.12g diff sol in H₂O Chlorine salts - wash gum and
3.0g Bismuth etc 4.86g Chlorine + 0.53g gum HNO₃ noticed in dist off $\therefore$ Chlorine salts present
easily sol in a small amt of cold ether & probably a mixture of various acids (K. discard).
(67°) Oxidation of a mixture of eq. anti l. threo & d. anthro gum from l. Xylene etc Jan 31-10

Took the 6.0g mixed of lactone gum [April 8-09 No. 35] called l. threo & gum (but present
later see No. 64-970 to be a mixture of l. threo & d. anthro gum) + mixed with 24.3g HNO₃ (1.21)
steated to 50° for 48 hrs then add the smoke up to 1L stake 10cc of 1.21 see $\frac{1}{2}$ N.H. needed.
Rest of aq. sol heated with 3.50g powder Ca (4 hrs Bismuth) gum 2.08g Ca Cro₄ + aq. filtrate crop
to a small volume gave no diff sol Ca salts $\therefore$ fllr by ale ale got 6.52g meso dried Ca salts
containing 17.3% Ca on analysis $\therefore$ had not of salts 6.3387g with 3.45g oxalic acid &
desalt off in vacuo 70° 25 min (HNO₃ noticed in d.f.) then taking up residue in warm
acetic acid got 3.35g gum (mobile & smelling faintly of HNO₃)
this took up 11.55g Chlorine & gum, in all,

(90)

a total of 7.35 gms (diff. sol in H₂O) Chinn salts - which with 5.0 g Brucine gave 6.1 g Chinn
d 1.07 g gum (Acetone extr). These gum with 5.8 g Brucine gave 7.15 g xx salts from xx these
from 20 cc absol. alcohol got 4.48 g crop I salts up to 120-220°; then crop xx from 3 cc H₂O
+ 75 cc alc (it don't dissolve in absol alc) gave 2 kinds of xx salt ∴ probably a mixture of
mess d 1,3 dioxyl glutaric Brucine discont all material at this point

(68) Arabinose + 33% NaOH final. Oct 7 09 Dec 28 09 Jan 22 10
Feb 19 10 Mar 15 10

251 g 99% NaOH in 400 cc H₂O (240°) flask quickly shaking a lot of
wool l. Arabinose, up to 150-160° in 200 cc H₂O + then local quickly pulling in Brucine
H₂O trap covered glass & then cool, dilute to 1123 g 20.16% HCl (theory 1115 g) & then
distill off in vacuum to x₂ e alcohol 200 g gum. This dissolved in 110 cc H₂O & extr 20 x by ether
gave 22.9 g other sol C₃ C₄ gums and 57 g gum (left in H₂O) which extr 2 x 1 x with 400 d 1 x
with 200 cc ethylacetate left a small amount of wood tar & NaOH. got from ethylacetate solutions
surprising over 1 x with H₂O 37.4 g gum; acetylated this 9 hrs 115 cc acetyl or 100 g got 59.45 g acetylated gums
of which all dissolved except 3.6 g tar in shaking with cold ether. add to ether sol. also the
acetylated gums obtained from 22.9 g C₃ C₄ gums (see below) & got a total of 56.5 g acetylated gums (from ether
sols) which by 2 x H₂O 3 x H₂O gave 36.7 g I + 35.7 g II then with 90 g Brucine 250 cc H₂O 100° alc gave
334.5 g C₃ C₄ C₅ gums. The 22.9 g gum ext from H₂O sol by ether as stated above was shaken
in cold 2 x with 200 cc ether & 1 x with 100 cc ether bearing with lime & formed 2.6 g wood berry gum
left 1.1 g insoluble tar (discont). got back from cold ether solutions 18.5 g C₃ C₄ gum S I
The 334.5 g gums was dissolved in 67 cc H₂O & extr 20 x by ether gave 8.3 g near soluble C₃ C₄ gums II
and head left in H₂O solution 25.4 g C₃ C₄ gums; shook the 8.3 g gum in cold with 200 cc ether
and all except 1.1 g by diff dissolved add to 25.4 g lot & det all as 26.7 g C₃ gums (2) $\alpha_D^{20} = 24.5^\circ (p=4.0)$
with 1.1 g insoluble tar (discont). got back from cold ether solutions 18.5 g C₃ C₄ gum S I

$\alpha_D^{20} (p=4.0) = -8.28^\circ$ ∴ 20% C₃ gums still present & libation of 1/25th ∴ 0.278 g look a total of 28.15 cc
w NaOH (16 cc in cold). The 18.5 g C₃ C₄ gums I treat with H₂O & KNC in Brucine, which off in new flask
up in alcohol etc gave back 16.6 g gum which gave $\alpha_D^{20} = -5.19^\circ$ ∴ 16% C₃ gums still present) & libation
of 1/50th (∴ 0.332 g) took total of 31.95 cc (NaOH 10.5 cc in cold). Now combine eq. solutions of C₃ C₄ gums
I and II & distill off in vacuum & dissolve gums in 50 cc H₂O & extr 20 x with ether & titration of 1/50th
gums after treating with H₂O KNC other 14.65 g C₃ C₄ gums (2) $\alpha_D^{20} = -2.047^\circ (p=4.0)$ & titration of 1/50th
∴ 0.293 g look total of 30.58 cc NaOH (12.5 cc in cold) ∴ treat rest of eq. sol. with 58.0 g Brucine
(theory 59.0 g) etc got 79.4 g xx salts which xx from 100 cc absol. alcohol & then got from filter 56.8 g salts which
gave 19.2 g crop I salts up to 185-180° $\alpha_D^{20} = -29.176^\circ (p=4.0)$ and then got from filter 56.8 g salts which
dissolved in 70 cc absol alc gave 17.44 g crop II up to 185-187° $\alpha_D^{20} (p=4.0) = -29.96^\circ$.
Now got back 34.6 g gummy salts which xx from 50 cc absol alcohol gave 10.5 g crop III all these Brucine
up to 206-208° (5.70 g) 113 g gum which was treated by digestion with 0.5 l ether into 27.8 g gum (see p 91)
Brucine and 1.13 g gum which was treated by digestion with 0.5 l ether into 27.8 g gum (see p 91)
into by 200 cc ethylacetate R.F. salts 1.13 g gum which was treated by digestion with 0.5 l ether into 27.8 g gum (see p 91)
20 g Brucine gave 26.6 g xx salts which xx from 60 cc absol. alcohol gave rapidly 14.08 g crop I
1,3 dioxyl glutaric Brucine up to 185-188° $\alpha_D^{20} = -30.63^\circ (p=4.0)$. The alc. filter. gave back 10.7 g salts
which by 50 g Brucine gave 7.25 g Brucine and 1.68 g gum (1.48 g white with 27.8 g above
on shaking this in cold with 50 cc ether all but 0.2 g gum dissolved & then
which was united with 0.22 g ∴ 0.45 g C₃ gums which
with 1.13 g above made 1.58 g C₃ gums (see p 92)

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(after 14.65g gum [X]²⁰ - 3.47g had been removed by other) (9.1)

The aq. solution left after extracting with ether gave 6.95g of C₅ gums [X]²⁰ = -14.84 (p=4.0)

was probably mixed with chinin (look up 20g) but gum on xx from a. w. d. no xx l. of d. time

of chinin is mixed alc. filtrate with a corresp. filtrate obtained from other 3 gum st

the 26.7g of gums (see p 90) found (p=4.0) to be [X]²⁰ = -24.5; titration taking two this is 0.267g

look 21.26cc $\frac{2}{10}$ NaOH (2.5cc in cold) solution for C₅ lactone = 20.22 cc. Now treat all of gum with 68.2g

chinin, then BurB for 3.31g insol. chinin from aq. filter got 109.9g xx salts which xx from 116cc

alcohol gave 28.39g d. then 3 chinin up 172°, crop I, the alc. filtrate gave 73.5g salts which xx

from 140cc alcohol gave only 2.6g more xx l. up 153-160° crop II. The alc. filtrate from crop II

was now combined with alc. solution of chinin salts obtained from 6.95g (14.84g) gum above

got back with 45g BurB alc 58.15g insol. chinin & 3x ether, 50.4g salts which xx

which when extr. R.F. with 0.5L ether look up 7.45g mobile gum which was resolved by shaking

with cold ether into 5.7g C₅ gums [X]²⁰ = -9.15 (see p. 90) + balance of 1.5th is 0.22g look

22.06cc $\frac{2}{10}$ NaOH (11cc in cold); and into 2.05g of gum just mentioned gave a total of

ether R.F. with alcohol alc including the 2.05g of gum just mentioned gave a total of

16.95g C₅ gums [X]²⁰ = -18.55 (p=4.0); titration of gum left in 0.5L

(for 3 lactone = 25.7cc) i. treat rest of aq. sol. with 49.6g Buerein alc for 74g xx salts which

xx from 100cc alcohol gave 22.3g crop I. l. another 3 Buerein up 175-177° when cold give

obtained from alc. filtrate got 20.25g crop II l. another 3 Buerein up 175-177° when cold give

more xx l. i. set for gum by 10g BurB for 15.53g Buerein & 6.6g black gum (alc extr)

which was i. extracted from 100-110° with 20cc only 10.8g alcohol from 10.8g salts which xx

remained wood on extr with other spr from other extractions 1.2g or 5.6g salts which by

15g BurB + 50cc H₂O this BurB to get back 5.25g of gum which look up in H₂O this BurB

11.8g chinin for 22.45g xx salts which on xx from 17.48g salts

chinin up 172° got from alc filter distilled off in vacuo & 4.35g of gum (cont tar) i. alcohol

salt i. set for gum by 10g BurB got 10.83g chinin & 0.9g tar remained wood on extr rather

with 14cc alcohol. this 10.8g got 6.9g alcohol from 3.95g then with 10g BurB + 25cc H₂O for

got on hydrolysis of other sol. gums 1x with H₂O → 3.95g then with 10g BurB + 25cc H₂O for

3.03g of gum (active other extr & 1x weep more) with [X]²⁰ = -9.52 (p=4.0). Now distill off H₂O

extr gum R.F. 2x with 300cc ether each time got 1.0g insoluble clear, almost colorless, gum

(w. vacuo) which was found p=4.0 to be absolutely inactive optically titration of 2.5 this

i. 0.32g look only a total of 20.83cc $\frac{2}{10}$ NaOH then for C₅ lactone = 19.76cc

as to C₅ lactone = 24.24cc

i. conclude that some synthetic C₅ gums must be found in traces from the

combination of these & these glucose molecules with each other O.E.L.

The rest of the C₅ gum extr by other 2.05g which must now be [X]²⁰ = -12.5

was now discarded.

New head (see bottom p. 90) following unidentified gums left
Other sol. gums $2.78g + 1.48g$ melted; and $1.13g + 0.45g$ 3-gums melted not sol. in cold ether.
The acc. del. of other sol. gums was gum $4.34g$ gum + titration of $\frac{15}{250}$ lbs (i.e. $0.2604g$) gave
 $26.99cc \frac{n}{10}$ NaOH :: heated rest of ag. sol. with $16.6g$ Borneol + got 1x sweep over c etc) $21.9g$ xx salts
which xx from 50cc alcohol gave $5.1g$ crop I 1,3 dioxyl butyric Borneol up to $185-190^\circ$ then
got $3.77g$ crop II up to $188-203^\circ$ obviously a mixture of lactide + 1,3 dioxyl butyric Borneol.
Meale. fill from crop I gum no more xx etc :: let fore gum by $6.0g$ Borneol + got $8.29g$ Borneol +
 $1.24g$ gum (other xtr) sol in 6cc cold ether :: heated it with $6.0g$ Borneol + got $0.76g$ gum + $0.76g$ Borneol +
Zinc + from ag. filtr adding etc etc got a total of $0.67g$ of lactate xx etc crop I only. ag. filtr
gum + gummy zinc salts only :: mix them & by salts :: discard.
1,3 dioxyl butyric lactone: Now took the various fractions of 1,3 dioxyl butyric Borneol as follows: $19.2g$ (old) = 2976
 $14.08g$ (old) = -30.63 ; $17.74g$ (old) = -29.46 :: total acid = $50.72g$ and free Borneol + all lactide Borneol
l. anthers & Borneol. got with $25.0g$ Borneol etc $39.0g$ red dry Borneol and $9.72g$ gum (other xtr) a
mobile acid smelling gum; the shaking this in cold with 50cc ether got a resolution into
5.62g other sol. gum - (see below)
and into 4.13g other insol gum (old) = -50.93 :: 3-anthracene present (16% or more); this gum heated with
4cc ph. $128-130^\circ$ was set 2 p = 4.0 absolutely 0,0 i.e. look 0.6935g xx and 16.6855g H₂O: p =
standing 40 hrs at 1011 and spread in 2 de lube = absolutely ± 0.0 Q.E.D. Keep rest of group.
the 5.62g other sol gum heated 7 hrs Borneol with 4.0g Zinc etc gave 1.01g insol Zinc +
got from ag. filtr one more adding etc etc a total of $0.47g$ of lactide of zinc only in 1st crop.
Now pple Zinc in ag. de filtrate and got $3.5g$ gum (other xtr). the heating this with 4cc ph. lye
+ 4cc alcohol at 7.44 hrs setting etc got $2.5g$ snow white 1,3 dioxyl butyric phenyl hydroxide
up to 130° (Keep) Q.E.D.
dl lactic acid. The 10.5g dl. lactic Borneol obtained, crop III, up to $206-208^\circ$ (see p. 90) gum with 5.0g Borneol
and 3.77g crops of Borneol salts up to $190-205^\circ$:: mixtures of 1,3 dioxyl butyric + lactide + got by 5.0g
Borneol etc 6.67g Borneol + $1.54g$ gum sol in 5pts cold ether :: melted with $17.6g$
lost: total = $3.3g$ gum other boiled with H₂O + $2.5g$ Zinc etc 7 hrs. Borneol gave $0.63g$ insol Zinc
+ ag. filtr cone + add etc etc gave a total of $3.03g$ crop II dl lactate of Zn 3 H₂O :: rest gum only, gummy
Zinc salts :: discard -
d. thms & Chlorin; Had a total of $29.78g$ salts (in 2 crops); on xx these from 240cc absol alcohol (R.F.) got
 $20.45g$ crop I d. thms salt up to 172° (old) = 103.93 i.e. look $0.9027g$ salt and $21.6452g$ H₂O: p = 4.002 +
of low melting acids. d. thms salt up to $153-160^\circ$ crop II p 91 and on xx these from 40cc alcohol got $4.6g$
more cp. d. thms Co salt up to 170° :: discard etc filtr containing about $7.5g$ salts only solubility
simply carried down mechanically by porous character of d. thms salt.
Now took the $25.05g$ cp. d. thms salt in 2 crops $20.45 + 4.6g$ + got with
 $12.0g$ Borneol etc $17g$ Chlorin and $6.93g$ (all xtr) mobile
colorless d. thms Co gum (old) = -36.5 (p = 4.0) d in 2cc
lube = -2.95 .

6.93g)

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Now make gum with H₂O up to 500cc + take 100cc $\therefore$ 1.386g lactone solut off H₂O in vacuum flask gum with 1.4cc ph. hydr. 57.4cc alcohol typ 0.7, 24 hrs seeking, adding acetic then till snow white 1.6g ph. hydr. up to 75-85° wash xx from 80cc hot acetic then gum 0.87g snow white hydrazide up to 108°-110° with $[\alpha]_D^{25} = +26.36$ i.e. 0.4635g sublt and 11.0865g H₂O $\therefore p = 4.015$ and found d in 10c tube = $+1.07$.

analysis: 0.2065g sublt. gum 0.4170g CO₂ and 0.1253g H₂O

Neely for C₁₁H₁₆N₂O₄ Found
C 55.00 55.05
H 6.67 6.74

Rotation Na salt took 50cc $\therefore$ 0.693g lactone gum and 52.5cc $\frac{2}{10}$ NaOH B.P.B. + distill H₂O in vacuo + desiccate Na salt, 0.9028g by theory, in H₂O to make total weight = 21.570g ($\therefore p = 4.18$) and found d in 2 de tube = $+2.01$ hence $[\alpha]_D^{25} = +23.76$

use rest of ap. gum 4.851g for acid expt see No 61

The l. anthero Co gum: had a total of 42.45g Bueim salt in 2 crops 20.25 + 22.2 which obviously consisted of a mixture of dl. 1,3 Dioxylutylene Bueim and l. anthero Co Bueim whose separation was exceeding difficult. The xx of the 42.45g salts from 300cc abs. alcohol got 31.2g crop I, salts up to 170°-172° with $[\alpha]_D^{25} = -36.5$; obviously much free Bueim present with salts; the alc filtrate from crop I gave 11.2g salts which xx from 30cc alcohol gave 5.02g crop II up to 160°-170° sale. filtrate from crop II gave 5.8g Bueim salts soluble in H₂O giving a neutral or very slightly alkaline reaction after heating on B.P.B. in H₂O solution: it is clear this part as probably being lacto Bueim.

Now took the crops I & II Bueim salt 31.2g + 5.02g salts = 36.22g + dissolved in about 250cc cold H₂O + 5x by C₆H₆ + proved that 25.2.35g up to 175-178 Bueim in C₆H₆ + filtrate got back 41.5g xx salts which xx from 50cc alcohol gave 0 because too little H₂O present: distill off 50cc at room temp 43.15g salts which on xx from 50cc alcohol gave on standing a large amt of heavy coarse grouped flat plates: add 50cc more alcohol & then filter typ 25.95g crop I up to dept 170-180° which was obviously a mixture of l. anthero Co Bueim and dl. 1,3 Dioxylutylene Bueim, and perhaps also some dl. lactic acid: had this with 13.0g BaCl₂ then C₆H₆ & then dry etc typ 18.85g Bueim and 5.42 mobile gum (Ether extr.) with $[\alpha]_D^{25} = -28.03$ ($p = 4.0$) i.e. found d in 2 de tube = -2.29 $\therefore$ about 75% l. anthero Co gum.

gum orig considered from l. anthero gum + acid by HNO₃ (Nos. 65 & 71).

The alc. filtrate from crop I above gave 8.4g salts which gave very little xx from 20cc alcohol $\therefore$ d & l free gum by 5.0g BaCl₂ etc typ 4.54g Bueim + 2.1g gum (ether extr.) with $[\alpha]_D^{25} = -23.87$ $p = 4.0$ and on getting rid of H₂O in vacuo + dissolving in 15cc H₂O + 20 by ether got from ag. sol 3.28g & gum (90° room) with $[\alpha]_D^{25} = -32.14$ ($p = 4.0$) and from ether solution got 4.25g mobile acid swelling gum which was now shaken in cold with 50cc ether and got in cold ether 1.22g gum with $[\alpha]_D^{25} = -14.22$ ($p = 4.0$) i.e. found d in 2 de tube = -1.15 ($p = 4.0$) $\therefore$ discard.

The portion used in 50cc cold ether was 2.8g gum with $[\alpha]_D^{25} = -25.84$ $p = 4.0$ $\therefore$ it is certain that this was 66% Co l. anthero gum besides the 3.28g which was about 80% l. anthero Co gum.

The 3.29g gum $[d]_D^{20} = -32.4$ was digested with H₂O + 8.1g Chinine 7hrs BarB & got 17.15g XX salts (after 2X sweep over c alcohol etc) & this XX from 30cc alcohol gave 2.6g, crop I, and d. thro G Chinine up to 160-165, & then in more XX the from alcohol.

Similarly the 2.8g gum $[d]_D^{20} = -25.84$ gave on digestion with 6.9g Chinine etc (1X sweep over c alc) 11.85g salts which XX from 25cc alc gave 1.85g d. thro G Chinine up to 150-153 & then got 9.25g salts which XX from 25cc alc and gave 2.46g crop II and d. thro G Chinine up to 150-155.

The combined alcoholic filtrates from the d. thro salts (both lots) about 12.65g Chinine salts gave back with 10.0g BarM₂ etc 8.92g Chinine & 3.0g gum (together extra) $[d]_D^{20} = -27.04$ (p=4.0)

which on shaking in cold with 50cc ether was most clear into 0.87g alcohol gum $[d]_D^{20} = -21.51$ (p=4.0) and 2 in 1 dc tube = -0.87

into 2.01g insol gum $[d]_D^{20} = -31.9$ (p=4.0) & in 1 dc tube = -1.29

The 2.01g gum was worked gently with 163.46 cc $\frac{n}{w}$ NaOH sol. remained very faintly alkaline & on dist. off H₂O in vacuo & dissolved Na salt 2.663g theory, so as to make total weight = 66.575g (i.e. p=4.0) and found d in 1 dc tube = -0.20 hence $[d]_D^{20} = -4.95$ which gives that ~~only~~ l. erythro G gum ~~must~~ be present as rotation of d. thro Na salt must be +23.76 over +12. $[d]_D^{20}$ total alcohol from Bruen salt $[d]_D^{20} = -18.77$ as J. Napier set free gum again from Na salt by 163.5 cc $\frac{n}{w}$ HCl etc got back 1.84g l. erythro gum which with 5.4g Bruen etc gave 8.54g XX Bruen salts which XX from 10cc alcohol. alcohol gave large amt of transparent flat square plates (incl on. amt of films needles) up to 170° crop I, 2.87g salts which XX from 10cc alcohol gave 2.46g XX the up to 180-182° $[d]_D^{20} = -30.72$ i.e. salt up to 180-185° which XX again from 30cc alcohol gave 2.46g XX the up to 180-182° $[d]_D^{20} = -27.67$ 0.9461g + 22.8404g H₂O i.e. p=3.977 & found d in 2 dc tube = -2.22

crop I 2.87g similarly XXX from 15cc alcohol gave 2.61g XX the up to 170° with $[d]_D^{20} = -30.72$ i.e. 0.9798g salt and 23.6631g H₂O i.e. p=3.984 & found d in 2 dc tube = -2.47

Napier obviously l. erythro G Bruen mainly containing in last case ~~some d. thro G gum free Bruen~~ permanent alk. sol. in warming on dist off H₂O in vacuo & dissolved Na salt 2.663g theory, in H₂O to total weight = 29.30g (i.e. p=4.0) found d in 1 dc tube = -0.20 i.e. $[d]_D^{20} = -4.95$ i.e. obviously l. erythro G gum present. Now set free gum by 75.6 cc $\frac{n}{w}$ HCl etc got back 0.87g gum which with 2.5g Bruen etc gave 3.87g XX salts whose $[d]_D^{20} = -29.09$ with $[d]_D^{20} = -29.09$

XX from 5cc alcohol slowly gave transparent flat plates & prisms. 1.69g up to 180-188° with $[d]_D^{20} = -29.09$ i.e. 0.9167g salt & 21.9699g H₂O (i.e. p=4.004) & found $[d]_D^{20}$ in 2 dc tube = -2.42 i.e. body was almost pure l. erythro G Bruen. The salt filter from crop I gave 1.72g salts which XX from 4.5cc alcohol gave 1.83-184° $[d]_D^{20} = -30.2$ i.e. mainly l. erythro G Bruen

gum 0.79g XX the up to 183-184° $[d]_D^{20} = -30.2$ i.e. mainly l. erythro G Bruen

The three crops of l. erythro G Bruen salts mentioned on top p. 94° 2.0g + 1.85g + 2.46g in all i.e. 6.31g were now reXX from 40cc alcohol and got only 2.42g salt up to 160° & this XX from 15cc alcohol gave 1.26g Chinine salt up to 165° i.e. still in pure d. thro salt

i.e. white all salts 6.31g again got with 4.0g BarM₂ etc 4.12g Chinine + 1.58g gum (ether expt)

Noticed now that the a. umm
0.28g xx from 3.5cc hot absol. alcohol gave 0.23g
melting at 160° it also xx in transp. tubular xx the see March 15 01 - 29,9
[2] $\frac{2.529}{2.049} = 29.08\%$ - 23.36 - 29,9
Now unite all Buein salts described on pps 94-95 as follows $\frac{2.529}{16.6g}$ Buein salts.
- 2.67g [2] $\frac{2.619}{11.74g}$ salts, mother liquors and in all a total of 16.6g Buein which should
2.46g ; 2.61g ; 11.74g salts, mother liquors and in all a total of 16.6g Buein which should
the total it was impossible to fit from 1. length to 3 Buein which should
all salts breaking in H₂O solution to combine all gums again
d. 25 mm 1 xx the salts of Cannel from
l. 1.36g

Now write all Brucine salts described in pps 94-95 as follows: Brucine salts:
 - 27.67 = 16.6g [27.67 - 11.07 = 16.6] $\therefore$ 16.6g Brucine salts
 2.46g; 2.61g $\therefore$ 11.74g salts, another liquor and in all a total of 16.6g Brucine which should
 find on further trial it was impossible to get pure l. erythro or Brucine which should
 melt up to 200°; but on melting all salts breaking in H₂O solution to combine all gum again
 with Brucine and then distilling off in vacuo 50° only at 25 mm & xx the salts obtained from
 5 cc H₂O 550 cc alcohol & scratch - as in case of separation of l. erythro & these gums from
 b. Xylene (see No 69) got 1st crop of xx salts which xx from 20 cc absol. alcohol gave
 on distilling off alc. at 50° and 25 mm got 120g salts which xx from 20 cc absol. alcohol gave
 5.43g, crop II, salts mpt 180°-183° with [27] = -33°53 $\therefore$ some pure Brucine present.
 Crop I, salts mpt 180°-183° with [27] = -33°53 $\therefore$ some pure Brucine present.
 Brucine & 0.92g stiff Co gum, which melted with 2.25g
 Chinin & H₂O xx to 3x with other & distill off H₂O in vacuo gave 3.48 xx salts easily soluble
 in 40 cc absol. alcohol & nothing comes out on standing $\therefore$ l. erythro & Chinin must be
 very sol. in absol. - in marked contrast to the d. erythro salt (see No 69) $\therefore$ can
 separate d. l. erythro easily into optical components by means of Chinine D. L. D.
 The 5.43g crop II gum with 2.5g Brucine etc 3.99 Brucine & 1.3g stiff Co gum which knew with
 H₂O & 3.19g Chinin Brucine then when cold x to 3x with other etc gave 4.5g xx Chinin salts which
 were easily soluble in 100 cc absol. alcohol & gave nothing on standing & scratching $\therefore$
 l. erythro & Chinin is very soluble in alcohol - D. L. D.
 Now get back gum from each lot by back 0.95g gum and 0.75g gums resp. (lost
 ether ex'ts). The smaller amt 0.75g gum with 0.8 cc ph. hyd. & 0.8 cc alcohol 0.7.38 hrs & etc
 other pot 0.95g l. erythro & ph. hyd. & 0.8 cc alcohol mpt 145°-150° 11

(from 100g l. xylise)

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29.35g G gums, which with 35.2g C₃ G gums about make a total of 64.55g Saccharins)
 with 0.52 ether left 6.0g gums insoluble are left w^h v^ol^um^e was found to

29.35g G gums, which with 35.2g C₃C₄ gums about 10.5cc in cold
The 35.2g C₃C₄ gums shaken in cold with 0.5L ether left 6.0g gum insoluble and left in vacuo
this gum after dissolving in H₂O, treating K₂C₂O₄, taking up in alc after dist off H₂O in vacuo was found to
weigh 5.5g, hence $\rho = 40$ $\Rightarrow$ $\rho = 19.4$ solution of $\frac{15}{250}$ (i.e. 0.33g) took 31.63 cc $\frac{n}{10}$ NaOH (theoretical
for lactone = 32.36 cc), $\therefore$ dissolved rest of recovered gum, 5.1g, in 11 cc H₂O + ext to 20 cc, ether
and head left in H₂O solution 2.2g C₃C₄ gums $\rho = 22.58$ ($\rho = 3.98$) whereas other sol problem,
10.5cc added to 26.15g left C₃C₄ gums (see below)
↓ cold ether was now ligated in H₂O sol. with K₂C₂O₄
back 26.15g C₃C₄ gums with 10.5cc in cold

$\alpha_7^m = +6.8$ ($p=4.0$) & got by titration of $\frac{1}{10}$ the 2.9g taken out by ether as stated above and
for extraction = 32.36 cc) $\therefore$ dissolved in 60 cc H₂O extra sox with other and got a resolution into
and had left in H₂O solution $\underline{2.2g}$ C₃C₄ gums (see below) $\alpha_7^m = +22.58$ ($p=4.0$) which
 $\underline{2.9g}$ by diff, was added to $\underline{26.15g}$ left C₃C₄ gums (see below)
The C₃C₄ gums dissolving in the 0.3 L. cold ether was now degated in the sol. with KNO₃
followed, distilled H₂O in excess, flake up residue in alk etc type back $\underline{26.15g}$ C₃C₄ gums with
 $\alpha_7^m = +22.58$ ($p=4.0$) which
 $\alpha_7^m = +16.7$ ($p=4.0$) which

$\alpha D_D^w = +7.8$
Now distill off H₂O from mixture
dissolved combined gums (about 28g) in 60 cc H₂O ($p=4.0$) and into 8.5 g of
 $\alpha D_D^w = +3.9$ and into 29.35g G gums II
18.7g other sol. C₃ C₄ gums with $\alpha D_D^w = +3.9$ and into 29.35g G gums I and
therefore makes 10.55g G gums I and 29.35g G gums II
Nee 29.35g G gums II give $\alpha D_D^w = +25.73$ ($p=4.0$) solution of tooth ($\therefore 0.2935g$) but 23.88 cc N₂ N₂H₄
heat time = 22.23 sec) $\therefore$ heated rest of aq. sol. with 76.6g Chinin thus BMB read 2.38g and thin.
135cc alcohol gave 16.55g crop I of L.H.M. d. rather G Chinin
toe filter from 1st gum I.

Dec 29, 35g & June
 (Heavy for G last time = 22.23 cc) :: March
 1st 135 gxx salts which dissolved in 135 cc distilled water xx from wood started about 60°
 up to 165°; the alc. filtr decolorized off pure 98.5 f salts which were white alc. filtr from 1st & June 5 f.
 up to 160° l. thru d. synthn of Chinine and then no more :: white alc. filtr from 1st & June 5 f.
 the 10.55g & June I pure with 25g Chinine B.P.B de a total of 38.55f salts on xx there from
 6.75g crop I, l. thru d. synthn of Chinine up to 160° when no more :: white
 2.55g + 4.45g filtr from G June II 1st with 50. - gas Burnt
 1.10g pure which was ext R.F. with 0.5 l ether
 a total of 7.95g aced at

the 10.55g of pure
50 cc alcohol for 6.75g crop I, 2.1.1.
alcoholic filtrate about 30g with comp. as 2.95g + 14.45g
type 61.65g Chitin + 2x 3x other + add for "ole" 2.35g pure which was ext. to 11.1.
taking out 9.95g mobile pure, of which we took 2x in cold with 100 cc ether a total of 7.95g
C₃ & C₄ gums dissolved with [27]₂₅^w = +5.44 (p = 4.0) solution of 10/250 th. = 0.318g loss 32.1 cc $\frac{22}{10}$ NaOH (19.5 cc in cold)
insoluble proteins (alc ext.) a total of 14.45g of gums with [27]₂₅^w = +21.76 (p = 4.0)
23.11 cc $\frac{22}{10}$ NaOH (5.3 cc in cold) (then for C₅ lactone = 21.70 cc).
1.15g pure with 42.5g Bucon and got 65.0g

C3 & found dissolved in insoluble proteins (see p. 10) Newm
 got finally also, from insoluble proteins took total of 23.11 cc $\frac{21}{10}$ Newm
 and titration of 50% (i.e. 0.289) took total of 23.11 cc $\frac{21}{10}$ Newm
 decided at this point (stupidly) to heat out of eq. sol 14.61 g from with 42.5 g benzene
 salts - which xx from woc alcohol gave 10.53 g crop I salts up to 150°-165° then from alc filtrate
 10.35 g salts which xx from alc woc gave 18.19 g crop II salts up to 175°-177° and then got from alc filtrate
 10.35 g salts which xx from alc woc gave 3.19 g crop III up to 170°-174°. The alc. filtrate gave 23.3 g salts which
 10.35 g salts which xx from alc woc gave 6.2 g brown gum which was acetylated then 100-110° with

48.35g salts which xx from 50cc alc fur $\rightarrow$ 3.11g
27.85g salts which xx from 50cc alc fur 15.9g Brucine and 6.2g brown gum when mixed with ether etc.
xx from fur with 10.0g BarH₂ etc 1.75g Tar remained insoluble on extn with ether etc.
20cc anhydrous & got 10.05g acetylated fur of which I c H₂O 5.6g then with 15g BarH₂ &
Hydrolysis of the oil acetylated fur gave 5.0g G gums & this treated
50cc H₂O & 1/2 lbs BwB got back

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with 12.0g Chinin etc gave 20.5g xx salts which on xx from 30cc alcohol only gave 0.25g l. then
d. extr. Cochinine up to 165° crop I; then 1.48g crop II up to 153°-160° then 2.05g crop III
up to 155°-160°.

The alc. filter from crop III now gave with 7.0g Ba₂ etc 8.0g Chinin & 3.6g from tarry i.
recrystallized it 9 hrs 100-110° with 11cc alcylid & got 0.87g tar & 4.53g other. sl. acetate from which on
hydr 1x the gave 2.95g from & then with 10g Ba₂ etc & see the other 100° gave 2.4g from with
[α]_D²⁰ = +13.1 (p = 4.0). The extr. then gave R.7a 2x with 300cc ether but left 0.73g
iron mobile stuff almost colorless gum with [α]_D²⁰ = +11.0 (p = 4.0) ∴ dis. curved but
obviously may consist of lactic & glycolic besides l. thion & d. nylthion & gums.

C₃ C₄ gums

Had 18.7g I [α]_D²⁰ = +03.9 and 7.95g II [α]_D²⁰ = +5.44.

18.7g lot, take 15 worth pt. 0.2805g of pure 30.58cc $\frac{1}{10}$ NaOH needed (26cc in cold) ∴ treat rest of eq. sol with
78.0g Buein & got xx of 106g salts from 130cc alc 17.87g crop I, up to 185°-188° with [α]_D²⁰ = -29.68 (p = 4.0)
then got 88g salts which xx from 100cc alc gave 23.95g crop II up to 185°-187° with [α]_D²⁰ = -30.16 (p = 4.0)
and then got 54g salts which xx from 70cc alc gave 19.03g crop III up to 207°-9° ∴ d. l. lactic Buein
plus got 34g Buein salts which with 18.0g Ba₂ etc. gave 22.7g Buein and 5.4 g gum which

4.3g easily dissolved in 100cc cold ether & rest 1.1g g gums remained in soluble limit with corresp. from
7.95g lot total with 30g Buein gave 37.35g salts which xx from 80cc alcohol gave

14.45g crop I [α]_D²⁰ = -29.4 (p = 4.0) ∴ mainly 1,3 dioxo butyric Buein other no more ∴ set free gum
from rest of salts by Ba₂ etc got 11.8g Buein and 2.95g gum [α]_D²⁰ = -29.4 (p = 4.0) which was
dissolved into 2.6g C₃ C₄ gums soluble easily in 50cc cold ether and into 0.35g C₃ gums left insoluble

with these with corresp. lts from 18.7g lot 4.3g & 1.1g ∴ have
6.4g C₃ C₄ gums undissolved and 1.45g C₃ gums ∴ the latter after degradation KNC etc gave back
1.38g gum with [α]_D²⁰ = +14.6 (p = 4.0) ∴ there is evidently another present as gum is too stiff for pure C₃ gum

∴ titration of this lot as det as 6.82g gum taking $\frac{10}{250}$ ∴ 0.2728g a total of 28.14 cc $\frac{1}{10}$ NaOH ∴ treated
rest of eq. sl. with 26.5g Buein & got 36.9g xx salts (1x weep on scale) got by loss in 70cc alcohol 5.52g
crop I salts up to 185°-190° ∴ mainly 1,3 dioxo butyric Buein; had a spill then flask cracking in
scratching crop I but got back 24.25g Buein salts which xxx from 30cc alc gave 5.05g crop II up to 188°

203° ∴ mixture lactic & 1,3 dioxo butyric Buein. The alc. filter from crop II gave 18.35g salts which xx
from 25cc alcohol was giving a fair amt of cubes of d. l. lactic Buein but decided now to set free gum from
all of the 18.35g salts by 7.0g Ba₂ etc & got 12.58g Buein and 2.62g gum easily sol in 12cc cold
ether. The gum was ∴ degraded with 2.0g zinc dust 0.54g metal zinc 3 & got on conc. eq. filter
solidly alcohol etc a total of 1.73 d. l. lactic & 3H₂O xx the besides some gummy zinc salts ∴ a
mixture of much lactic & 1,3 dioxo butyric lactone.

2 de tube was -0.26 hence $[2]_D^{20} = -3.15$. According to calculation l. then G Na should give $[2]_D^{20} = -23.76$ and d. anthers G Na should give $[2]_D^{20} = +15$ to $+20$ ∴ about eq. amts of 2 gums must be present - a conclusion also reached from rotation of ^{mixed} Chlorine salt G.E.D.

Now get back gum l. then d. anthers from rot of eq. sol (840cc).

The xx Buein salts obtained from the 14.45g G gum $[2]_D^{20} = +21.76$ p. 97 bottom

Had a total of 31.84g salts in 3 crops (18.1 + 10.55g + 3.19) supposed for a long time to be the salt of the l. then G gum which is n.g. as proved later. It is obvious that these salts represent mixtures of

the l. then d. anthers G gum with more or less 13 droxy butyric present.

See xx the 31.93g salts from 200cc alcohol got 21.25g crop I mpt 170-172° with $[2]_D^{20} (p=4.0) = -30.5$; got from filtrate mxx from 200cc alc 5.81g crop II mpt 160-170° (sale. filtr. was discarded).

Now dissolved crops I + II salts in about 150cc cold H₂O & H₂ which removed by dec. det

1.73g Buein mpt 175-178° from distilling off H₂O in vacuo for 29.7g xx salts which xx from 50cc alc gum

0 as too much H₂O present but on distilling alc again in vacuo & heating salts 28.45g with 50cc alc alcohol

then heat on B₂ till almost dissolved

not all would dissolve but have much insol mxx powder in heat on B₂ till almost dissolved

then cooling for large amount of heavy thin flat plates with pointed ends 18.8g crop I mpt 170-175°

crop I gum with 10g B₂H₄ 13.25g Buein and 3.93g gum [Ether extr] with $[2]_D^{20} = +20.4 (p=4.0)$

∴ clearly a mixture of about eq parts G & O gums -

The alc. filtrate from crop I had 7.45g salts which with 5.0g B₂H₄ gave 3.9g Buein and

1.72g gum (Ether extr) with $[2]_D^{20} = +29.67$!!

Now took 4 similar fractions called d. anthers G Buein from l. Xylose expt No 35 p. 50 of June 22-09

Took 3.07g salt mpt 194° $[2]_D^{20} = -25.1$ and 1.61g salt mpt 192° $[2]_D^{20} = -25.64$ and 3.0g salt $[2]_D^{20} = -26.85$

mpt 170-177° from 4.1g for my used 11.6g for rotation and 5.84g mpt 185-190°

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and then got 2.08g exp^t salt repr 150-155. Unite these: 10.85g + xx from 65cc abs^l alc
and got 6.46g l. thos d. anthers Chimim repr 163-5 with $\alpha D_D^w = -115^{\circ}3$ $p = 4.0$ $\therefore$ l. thos predominates
 $\downarrow$ sec. fols discarded.
The orig. sec. filtrate from exp I + II thus anthers Chimim gave back 7.7g salts which with 5.0g Brucine
we gave 4.53g Chimim + 1.51g gum (either extra) with $\alpha D_D^w = +150^{\circ}57$ ($p = 4.0$) $\therefore$ unite
this with 1.0g gum $\alpha D_D^w = +10^{\circ}88$ see p 100 near bottom $\therefore$ 2.5g undantifol gum + shock
this we cold with 50cc ether + resolved it thus into 1.14g ether sol gum $\alpha D_D^w = +6^{\circ}18$ ($p = 4.0$)
and into 1.35g unsol gum $\alpha D_D^w = +19^{\circ}53$ ($4.0 = p$). The 1.14g l^r braled with 10.8cc $\frac{N}{10}$ NaOH
gave a faintly alk. sol after warming + on d. of H₂O in vacuo + making up Na salt 1.572g by theory
to total weight $p = 4.0$ $\therefore$ found d in 1 cc tube was -0.087 whence $\alpha D_D^w = -2^{\circ}15$
Similarly the 1.35g gum braled with 117.3cc $\frac{N}{10}$ NaOH till just faintly but perm. alkaline when warm sol
gave on d. of H₂O + making up Na salt 1.819g by theory, with H₂O to give bral weight ($p = 4.0$)
gum αD_D^w in a 1 cc tube = -0.293 whence $\alpha D_D^w = -7^{\circ}25$. These results simply prove that
l. thos of Na salt is present in slightly larger ratio than d. anthers of Na Lef p 100 (p. 99 bottom)
Proof that the Chimim salt repr 166 $\alpha D_D^w = -113^{\circ}2$ + its resulting gum 10.7g p. 99 bottom
was a mixture of 9. pts l. thos + d. anthers of ^{metas} ~~saccharine~~ salts -
(Had 7.918g l. thos d. anthers gum left from the 10.7g gum ^{used} 260cc for expts, the
last 100cc being used with Brucine Jan 22-10 + is not described in detail here.
 $\downarrow$ Then used also 3.5g l. thos d. anthers gum obtained from various sources of d. thos l. anthers
Chimim repr 163-166 eq. the 6.46g above; the 3.32g obt from xx of 6.04g crude salt
repr 160-165 Dec 29-09 + also 2.33g + 1.33g sp. l. thos d. anthers salt May 12 09
 $\therefore$ got in all about 11.4g l. thos d. anthers to gum, for separation into components.
These were now braled with 34.0g Brucine + gum 5.2g xx Brucine salts (see l^r next page)
A prelim expt with 1.07g d. thos d. anthers gum + 3.0g Brucine had established that got a resolution of 4.3g xx
into d. anthers of Brucine by xx from 10cc alcohol as 1st crop obtained by xx from 10cc abs^l alc gave
2.96g xx repr 185-190 where when xxx from 70cc alc. gave 1.42g xx previous repr 185-190 with
 $\alpha D_D^w = -25^{\circ}64$ ($p = 4.013$). Subsequently noticed that this salt comes out very slowly from alc cont
H₂O solⁿ repr up to 202^o or higher and αD_D^w goes down to $-22^{\circ}75$ or even lower. The
l. thos salt is not carried down to as large extent under these conditions as when ^{about} abs^l alc alone
is used when simply always get mixed xx l^r of the salts of the 2 acids - even at
first resolution (a quant^y is excruciatingly diff shes not yet been accomplished this)
both products have been obtained pure -

The 11.4g d. anthro l. thro from pure with 34g Buem ale 5.2g salts which dissolved in 60cc alcohol
 & let stand (seeding with crude d. anthro Buem up to 185-193° & $\Delta T_D = -25.64$) gave 21.62g salts up to
 160°-180°: a mixture; the alc. filtr gave 28.0g salts which xx from 100 cc hot alcohol, alcohol
 gave 11.88g crop II salts up to 185-192°: mainly d. anthro salt
 crop I rexx from 300 cc alcohol gave 6.94 salts up to 180°-190° and then 5.24g salts up to 185-187°

After much work got a separation into 23.78g salts melting from 180°-194° and into an
 alcoholic filtrate containing 18.65g (b) gums very easily sol. in alcohol salts
 the former (a) could not be got pure by repeated xx from alcohol large or small amounts & is
 a mixture of much d. anthro salt containing l. thro salt in varying proportions.
 Now took (a) 23.78g salts, dissolve in H₂O, extr 5x C₆H₆ (to remove supposed excess of Buem) etc
 got distilling off H₂O at 50° only at 15 mm got 26.05g salts which dissolved in 75 cc alcohol gave
 1st bulky opaque nodules in large amounts & then slowly on top of these got heavy xx prisms
 1st heavy xx prisms 7.65g xx prisms up to 190°-200°-obviously mainly d. anthro & Buem;
 easily on gentle warming: got 5.72g d. anthro & Buem up to 202°-3°

On xx this from 4.5 cc H₂O and 50 cc alcohol this gave 5.72g d. anthro & Buem up to 202°-3°
 with $\Delta T_D = -22.71$; i.e. 0.8321g salt and 20.0404g H₂O: $\beta = 3.987$ found Δ in 200 tube = -1.83.
 It is doubtful if this is cp. d. anthro Buem - but can use this for the purpose. Now set free gum
 from this with Chlorine (or better still probably by conversion into the ph. hydrazide). Now set free gum
 from all of 5.72g d. anthro salt by 3.0g Buem + got 4.15g Buem and 1.18g gum (but other extr)
 with $\Delta T_D = +54.08$ ($\beta = 4.0$) (Δ in 100 tube 2022). N.B. a small amount of gum is always left in the soil in
 acetone, etc or acetone other & this is less + hence got (a) usually about +40°.

The 1.18g +54.08 gum from 2.89g Chlorine ale 4.65g xx Chlorine salt which is very diff. sol. in alcohol
 alcohol & gums are very completely on cooling; the 4.65g salt eq. took 80 cc alcohol to dissolve and
 on standing got 3.05g xx needles up to 172° with $\Delta T_D = -104.0$ i.e. 1.0119g and 24.3090g H₂O ($\beta =$
 3.996) gum Δ in 100 tube = -4.20; the salt can also be xx from a small amount of H₂O. The alc. filtrate from

3.05g crop I gum in cone 0.43g more crop II same salt: 172°: water (see p 103)
 Now took (b) 18.65g salts added Buem xtr by C₆H₆ from (a) above to it, heated in H₂O solution, cool & xtr
 5x C₆H₆ etc got distilling off H₂O only at 50° at 25 mm 21.35g xx salts which xx from 50 cc alcohol
 slowly gave opaque bulky xx nodules, 10.94g crop I up to 150°-160° (fuses xx H₂O); on xx these from
 5.5 cc H₂O & 60 cc alcohol got slowly cone. grouped flat prisms, 4.62g, up to 145°-150° crop I
 l. thro & Buem with $\Delta T_D = -34.07$ mix this with 1.33g salt $\Delta T_D = -33.47$ took up for
 d. thro gum (see p. 103). The alc. filtrate from crop I l. thro salt gum in cone ale 3.0g xx ale up to 145°-150°
 with $\Delta T_D = -27.80$: a mixture of l. thro & d. anthro (eq. p 103).
 The orig. alc. filtrate from crop I, 10.94g above gave on distilling off alcohol & less salts (about 11g) in 200 cc
 alcohol 3.43g xx nodules up to 170-183°: mixture.

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{ 3.6g C₂ gum other extn
2.6g C₂ gum (stiff) also extn }

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Without going into further details the Brucein salts (a) and (b) were separated further by
xx various fractions acc. to mpt. pt. rotation etc, into 6.78g more d. anthers Brucein
mpt 190°-202° and into another 1.33g xx fraction. l. then salt up 145-150° with 10.0g d. anthers 14.9g
whereas rest still remained mixtures of 2 salts in question (these gum with 10.0g d. anthers 14.9g
l. then C₂ gum; Now took the 2 fractions of l. then salt (pp 102-103) 4.85g xx Brucein and lost other extn 1.11g
-33°47' ∴ l. then = 5.95g salts; got with 3.0g BaCl₂ 4.24g Brucein and lost other extn 1.11g
l. then gum [α]_D²⁰ = +42°5' (p = 4.0) (V.B. trace of C₂ gums not extn by other in all cases)
Now treated the 1.11g gum with 2.72g Chinin typ 4.85g xx Chinin salt (l. then) which xx from 30cc alcohol
gum 2.14g xx needles mpt. 160°-2° with [α]_D²⁰ = -119°45' (two lines as salt not abso. free from d. anthers).
1.2 0.960g salt and 22.989g H₂O (p = 4.007) gum d in 1 dc tube = -4°8'4; the alc. filtrate lost off +xx from
10cc alc. gum 0.50g mpt. 160°-2° ∴ l. then salt and especially also the d. anthers Chinin salt ∴ l. then gum is vertically more
soluble in alcohol than d. then salt and especially also the d. anthers Chinin salt ∴ l. then gum is vertically more
d. anthers C₂ gum II; the 6.78g crude anthers gave with 3.0g BaCl₂ 4.92g Brucein and
1.4g d. anthers gum + this with 3.4g Chinin typ 4.75g xx salts which xx
from 70 cc alcohol gave a total of 3.33g d. anthers Chinin salt up 179° with [α]_D²⁰ = -105°7
1.2 0.919g salt and 29.8874g H₂O (p = 2.983) too much H₂O by mistake) found d in 1 dc tube = -3°8'.
The alc. filtrate lost off +xx from 10cc alcohol gave 0.47g more Chinin salt up 171° with [α]_D²⁰ = -3°8'.
Now took the various d. anthers Chinin salts as follows 3.33g + 0.47g + 3.05g + 0.43g (p 102)
∴ about 7.0g treated with 3.0g BaCl₂ etc typ 4.95g Chinin and 1.93g gum (lost extn); now divided
this into 2/3 and 1/3 (use 1/3 for ph. hyd + 2/3 for Brucein salt [seep])
ph. hydrazine 1/3 of the 1.93g gum treated with 0.65cc ph. hyd + 0.6cc alcohol greatly gave xx needles on standing
∴ after 24 hrs add acetic ether, filter wash with typ 0.88g mpt 140-145° very diff sol in ethylacetate even on boiling
∴ lost use alcohol for xxx. Before filtering add 10cc ethylacetate, shake then filter, wash c. cold ethyl-
acetate - got 0.68g ap. d. anthers C₂ ph. hydrazine snow white mpt 150° with [α]_D²⁰ = +9°38' i.e.
0.3971g salt and 9.6477g H₂O (p = 3.753) and found [α]_D²⁰ in 1 dc tube = +0.375. acid ph. hyd from 29.5g salt (see 10 68 p)
analysis 0.2076g salt gave 0.4202g CO₂ and 0.1272g H₂O
Found
C 55.00
H 6.67
N 3.81
Theory for C₁₁H₁₆N₂O₄
C 55.00
H 6.67
N 3.81
l. then C₂ gum continued: took the 2.64g l. then C₂ Chinin salt typ with 1.2g BaCl₂ etc 1.85g Chinin
and 0.8g gum (lost extn); this with 0.8cc ph. hyd + 0.8cc alcohol got entirely solid in 20 hrs. O.T. after 24 hrs
add acetic ether, stir filter wash typ 1.1g and a hydrazine mpt 100-110°. This xx from 100 90cc
lost acetic ether gave 0.79g ap. ph. hyd mpt 110-120° as a snow white, very
bulky ppt. [α]_D²⁰ = -25°44', i.e. 0.3721g salt and
9.0081g H₂O ∴ p = 3.966 gum d in

get back phlyd + (keep)

1 de tube = -1,02.

Analysis.

0.1953g subst gnr 0.3948g Co₂ and 0.1186g H₂O
0.2069g " gnr 21.6cbm by N₂ at 22°25' and 730 mm

Heavy for C₁₁H₁₆N₂O₄

C	53.00
H	6.67
N	11.67

Found
55.12
6.76
11.65

dl thru to ph. hydrozide see No. 68 p. 6
d. anther of Buein (cont from p. 103)

Work 0.1g of above c.p. l. thru phlyd + 0.1g red sp. d. thru phlyd
see schizoid in anant H₂O drops in desicc in vials of powder
melt 130° with softening before begin at 120°: xx it from 1cc about alc
+ some a solid cake of bulky needles: add 5cc ethyl acetate yellow
0.14g xx lte up to 120-130° (keep): not measure.

Took the 3/4 1.93g gum + treated Buein with 3.8g Buein
5.4g xx salt which xx from 1.2cc H₂O + 25cc alcohol gave
very slowly transparent cone grouped 4 sided flat prisms in large amt, which gradually
grew to very heavy opaque nodules: got 4 days at 170° with xx H₂O of
ethyl solid then melts towards 200° again. The salt loses xx water very slowly and in
small amounts in completely at 100°-105° + in trying to drive xx H₂O off by heating higher
found that slight decomposition took place: can't remove xx H₂O. The slightly decomposed
salt was: rexx from 2.17cc H₂O + 20cc alcohol, after adding 2.17g of a similar by one

obtained later from 2cc H₂O + 20cc alcohol and got 4.96g salt
salt (-23°05' [α]_D²⁰) showing that main portion of salt had been
melt 200°-202° with previous softening at 170° showing that main portion of salt had been
converted into the only dried salt by this process: the salt, 4.96g, gave [α]_D²⁰ = -21°83
p = 4012 i.e. 1.0131g salt and 24.2482g H₂O and found 2 in 2cc tube = -1°77.

Now used all this lot 4.96g for anal. to mass choxy glutaric acid see No 66 p 89.
Non identified G gums: as mentioned on p. 103 top, got fine rest of Buein salts by Buein
etc 3.6g G gums (other extr) and 2.6g G gums (alc extr). As it was at first thought it might
be possible to separate d. anther + l. anther G gums - owing to diff in sol. of their chimer salts
in alcohol - it was necessary to prove that this is an impossibility. Accordingly foll. expts
were carried out. The 3.6g lot work in H₂O sol 8.85 Chimer + after 3x extr c other etc got 12.85g
salts which xx from 70cc alcohol, alc. gave 6.7g salt, gave 5.0g salts which xx from 15cc alcohol gave

(p = 4.0) on rexx this from 90cc alcohol etc seeding with ep. d. anther Chimer up to 162-4° [α]_D²⁰ = -111°8
got 4.68g salt, gave 3.21g xx lte up to 168°-170° [α]_D²⁰ = -111°5: can't separate l. anther from d. anther Chimer
The orig. alc. filtrate from cup I 6.7g salts, gave 5.0g salts which xx from 15cc alcohol gave
2.04g cup II up to 160° [α]_D²⁰ = -111°9.
dodo, on treating the 2.6g gum with 6.5g Chimer etc 3x extr if got from 50cc alcohol 2.34g xx
up to 160°-20° with [α]_D²⁰ = -109°1 and then 2.22g cup II [α]_D²⁰ = -108°5 up to 160°
Finally took a total of 8.0g Chimer salts where [α]_D²⁰ ranged from 107°4 to 113°2 started with 4.0g
Buein etc got 5.23g Chimer + 2.26g G gums; these treated
with 6.75g Buein etc gave 9.55g xx salts

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which xx from 40 cc absol. alcohol gave 3.75g salts, crop 7, mpt 160° - 170° gets solid when melts $195-200^{\circ}$ $\therefore$ mainly hydrated d. anthers & B. in as $\alpha 7^w$ found to be $-23^{\circ}03$.
also 2.95g gum from easily sol. Chinin salts left in filtrate & these with 8.85g B. in
extr 3x $\frac{1}{2}$ H₂O etc gave 11.5g salts which xx from 20 cc alcohol gave 4.0g xx the mpt 160°
& these when xx from 1 cc H₂O + 20 cc alc gave 1.39g crude l. then B. in xx the mpt $155-157^{\circ}$
with $\alpha 7^w = -31^{\circ}96$.

As a final outcome stopped when had a total of about 16.4g B. in salts - not yet
identified ^{1.2} resolved into fairly pure l. then or d. anthers salt as shown by mpt
& polar is cope - these were $\therefore$ added to B. in salts and x from a new lot
of l. then d. anthers gum see Vo 70

(70) l. Xylose + 33% NaOH May 19-1910

2 lots of 100g l. Xylose ^{mpt 150-50} each acc to 6g (initial temp + 50 $^{\circ}$ next temp + 80 $^{\circ}$): used 251g 787%
NaOH + 1123g 2016% H₂O theory 1114 (each can). The eq. dist ^{combined} took in all 220 cc $\frac{w}{w}$ NaOH
(ph. ph. rate 6 drops: not over $\frac{1}{2}$ cc each added). Now distill off H₂O = 600 cc; & distill off 600 cc from
this; and then from the 600 cc + 1 $\frac{1}{2}$ 5 cc sat. c. H₂CO₃ gum less than 0.3 cc each off (from ph. ph.) $\therefore$ not
CH₃COH can be present as had been found before see

The salts residues extr each 2x with 2 L ether each time gave 49.7g & 51.0g gum resp -
which combined & dissolved in 100 cc H₂O & extr 20x by ether gave 59.1g C₃ gums $\alpha 7^w = +4.0$
p = 4.0 and got from H₂O solution 31.65g G gums. A $\alpha 7^w = +24.0$ p = 4.0
the extr. salt residues in 4 lots 1x more with 1 L ether each got 16.45g G gums B $\alpha 7^w = +21.7$ (p = 4.0)
Now extr salt residues all with cold alcohol till snow white & get much G gum C containing last
NaOH $\therefore$ extr these again 5x R. 7 with 1 L ether each time and got 14.8g G gums C with $\alpha 7^w$
+28.0 and +30.0 (could not get sharp reading p = 4.0). The residue now left weighed 34.0g so x 6 it
R. 7 with 250 cc ethyl acetate got 12.2g G gums D $\alpha 7^w = \text{over} +33^{\circ}$ (p = 4.0) leaving $\therefore$ 21.8g ^{l. then d. anthers Chinin}
invol NaOH and tar (discarded). Gums ^{xx from} A took up 88.5g Chinin + gum 31.87g l. then d. anthers Chinin ^{mpt 160-5}
Gum B took up 43.0g Chinin + gum 18.45g l. then d. anthers Chinin ^{mpt 160-5}
Gum C " " 40.0 " " 21.42g l. " " " " " " "
Gum D " " 32.0 " " 25.50g l. " " " " " " "

gave 6732g Chinin + 24.55g l. then d. anthers salt from A B C + D = 97.24g which with 45.0g B. in
etc gave 110g salts (1x sweep over) & on drying this with 600 cc alcohol breaking up till all
a granular powder much dissolved on cooling got 53.9g xx the mpt 170° - 190° $\alpha 7^w = -27^{\circ}3$ (p = 4.0)
 $\therefore$ about 600g anthers + 40% then salts. These xx from 20 cc H₂O + 250 cc alcohol seeing with
d. anthers salt mpt 20° ; gave rapidly in 24 hrs at 21.21g salts mpt 190° - 200° with
 $\alpha 7^w = -24.9$ (p = 4.0) and these xx from 10 cc H₂O + 100 cc alcohol
gave 16.35g salts $\alpha 7^w = -23^{\circ}2$

which together with 4.96 g salt - 21.81 was used in ~~acid~~ expt No 66 p 89
Have all rest of Burein salts ~~from No 69~~ ^{also 16.48 Burein salts from No 69} in 3 fractions acc. sep. of l. three from
d. synth in fall + to get amounts of each body accurately from ap. ph hydrazines etc.
i.e. kintal as given ~~45.85g~~ ^{46.00g} salts which xx from 200 cc alcohol gave
45.85g salts $[d]_D^{20} = -27.06$ up to $155^{\circ}-160^{\circ}$; the alc. filler gave 37g salts usual xx from
50 cc alcohol gave 13.23g salts up to $170^{\circ}-175^{\circ}$; $\therefore$ unite these 58.08g ~~58.08g~~ ^{58.08g} ~~58.08g~~ ^{58.08g} from 22 cc tho
+ 200 cc alcohol ~~gave~~ ^{gave} 22.9g xxlle up to 160° $[d]_D^{20} = -29.24$ [keep] ~~thru Oct. 1, 10~~ ^{thru Oct. 1, 10}
The alc. filler cont.: 35.18g salts was dissolved in 80 cc alcohol vis now in d.f. full of xxlle (keep)
3rd fraction is in a flask on shelf - about 22.0g salts [keep] ~~thru Oct. 1~~ ^{thru Oct. 1}
The orig. de. solutions obt. as follows from l. thro d. synth in Chinix xxlle from A B & C ~~thru Oct. 1-10~~
still contained about 180g Chinix salts - since they gave no more xxlle from alcohol, now set free
gave by 90g BaCl₂ ~~thru Oct. 1~~ ^{thru Oct. 1} 145.7g Chinix + 44.9g mobile gum (alc extn) $[d]_D^{20} = +4.07$ (p=4.0)
This gum dissolved in 50 cc H₂O + extn 20x by ether gave 17.0g C₃ gum $[d]_D^{20} = +4.07$ (p=4.0)
1 cc of this with 1 cc ph hyd + 1 cc alc gave only 0.1g 13 dux + 1 cc ph hyd r. $\therefore$ disc all
and 24.5g C₃ gum $[d]_D^{20} = +22.03$ (p=4.0) ~~thru Oct. 1~~ ^{thru Oct. 1} 31.95g ~~thru Oct. 1~~ ^{thru Oct. 1} 5.75g C₃ gum $[d]_D^{20} = +9.36$
over cate) + on xx from 200 cc alc ~~thru Oct. 1~~ ^{thru Oct. 1} 31.95g ~~thru Oct. 1~~ ^{thru Oct. 1} 5.75g C₃ gum $[d]_D^{20} = +9.36$
with $[d]_D^{20} = -113.01$ (p=4.0) [keep] ~~thru Oct. 1~~ ^{thru Oct. 1} 5.75g C₃ gum $[d]_D^{20} = +9.36$
BaCl₂ etc ~~thru Oct. 1~~ ^{thru Oct. 1} 36.0g Chinix and 14.85g gum (alc extn) $[d]_D^{20} = +14.83$ ~~thru Oct. 1~~ ^{thru Oct. 1} 5.75g C₃ gum $[d]_D^{20} = +9.36$
14.85g gum in 15 cc H₂O + extn 20x by ether was dissolved with 5.75g C₃ gum $[d]_D^{20} = +14.83$ ~~thru Oct. 1~~ ^{thru Oct. 1} 5.75g C₃ gum $[d]_D^{20} = +9.36$
 $\therefore$ much C₃ gum (25% present) and into 8.4g C₃ gum $[d]_D^{20} = +14.83$ ~~thru Oct. 1~~ ^{thru Oct. 1} 5.75g C₃ gum $[d]_D^{20} = +9.36$
is obviously l. thro d. synth from mainly - the taking 1 cc of 5.75g gum + 1 cc ph. hyd + 1 cc alc
gave 0.36 lbs de only 0.07g 13 dux + 1 cc ph hyd r. up to $125^{\circ}-130^{\circ}$ $\therefore$ disc all
(71) Formose, 33% NaH. ~~thru Oct. 1~~ ^{thru Oct. 1} 31.95g ~~thru Oct. 1~~ ^{thru Oct. 1} 5.75g C₃ gum $[d]_D^{20} = +9.36$

(71) Formose, 33% NaH. ~~thru Oct. 1~~ ^{thru Oct. 1} 31.95g ~~thru Oct. 1~~ ^{thru Oct. 1} 5.75g C₃ gum $[d]_D^{20} = +9.36$
Formose was made from 347g cat. K₂S₂O₈ 35% H₂O + 2984g H₂O + 14.6g PbO₂ heating 3.5 hrs very hard
shaking frequently for 5 hrs then add 7.65g C₃ gum H₂O made ~~thru Oct. 1~~ ^{thru Oct. 1} 17.5g PbO₂ (heavy 17.87g) ~~thru Oct. 1~~ ^{thru Oct. 1} 17.5g PbO₂ (heavy 17.87g) ~~thru Oct. 1~~ ^{thru Oct. 1} 17.5g PbO₂ (heavy 17.87g)
110g crude Formose; Now ~~thru Oct. 1~~ ^{thru Oct. 1} 275g 98.5% NaH in 450 cc H₂O (+38") shake with the
formose diluted with 200 cc H₂O + 55 cc more H₂O (2L flask as usual) + then heat Boil 8 hrs
(H₂O has de) ~~thru Oct. 1~~ ^{thru Oct. 1} 1230g 20.2% H₂O (heavy 12.21g) ~~thru Oct. 1~~ ^{thru Oct. 1} 1230g 20.2% H₂O (heavy 12.21g) ~~thru Oct. 1~~ ^{thru Oct. 1} 1230g 20.2% H₂O (heavy 12.21g)
105g gum; on extn this R.F. 3x with 1 L ether got 34.0g very mobile gum; since the problem left
unsol in ether was very heavy 67g ac. sol, it was acetylated with 200 cc anhyd r. ~~thru Oct. 1~~ ^{thru Oct. 1} 104g g acetylated
fund of which 26.0g tar + NaCl remained insol on extn with ether; the acetylated gums sol in ether,
70.85g, were ~~thru Oct. 1~~ ^{thru Oct. 1} 102g gum B (alc extn); these acetylated gums were 101.65g A and
104.85g gum A and 102g gum B (alc extn); these acetylated gums were 101.65g A and
gave 47.15g + 44.80g tar + NaCl mainly, used in ether. The other sol. acetylated gums were 101.65g A and
103.0g B; A " " 2nd " with 130g BaCl₂ + 300 cc H₂O 59.8g gum A + B gum with 130g BaCl₂ + 300 cc H₂O 59.8g gum
A + B were now each extn 3x R.F. with 1.0L ether each time ~~thru Oct. 1~~ ^{thru Oct. 1} 59.8g gum A + B gum with 130g BaCl₂ + 300 cc H₂O 59.8g gum
43.15g soluble gum from A and 40.1g sol. gums from B
got insol gum 18.33g A and 21.15g B

gums A B + C since they titrated mainly for C_6 saccharins were treated with
Styehrine etc to get a solution into the dextro- & galactose-metaseach. Styehrine salt
which should come out from alcohol as xx salts - but got oils only $\therefore$ n.g. $\therefore$ Set
free gums again & got back ^{Booth's} 10.05g A ; 14.45g gum B; and 8.88g C.
A was now treated with 24.4g Buein and gum in all 10.48g Buein salts exp I upr 180-188° when
3.95g exp II upr 170-180°
B was treated with 35.15g Buein + gum by xx in all 5.3g Buein salt upr 195-202°
C ^{incl 4.24g (see below)} 23.97g ^{xx Buein salts (upr 170-202°)} 21.60 " + gum by xx " " 0.05 Buein salts; had a total of
back with 40.0g BaM₂ etc got 61.2g Buein and 2.30g (2x) gums; the extra said we
6x with 800cc ether (R.F.) dissolved 7.35g (2x) ; 5.79g (2x) and 4.6g more of gums } see p 109
1x with 800cc acetic ether for 6.77g gums and then with ale for 2.4g gum (the extra) $\therefore$ 7.0g of gums
Gum C gum back 15.7g Buein + 6.28g gum sol in acetic ether and 2.4g gum (the extra) $\therefore$ 7.0g of gums
C4 gums: Had 28.8g C₃ gums (a) and 7.05g (b) and 9.05g (c) which last two were mixed $\therefore$
16.1g gum. The 28.8g lost taking 100 to the tank 27.50 cc $\therefore$ NaOH: labeled rest with 107g Buein + gum
all in 3 crops (44.35g upr 144-145°; 24.25g upr 185-192°; 7.49g upr 180-187°) a total of 76.09g gum
1,3 Divy butyric Buein; the ale filtrate from 1,3 Divy butyric Buein and 6.43g Buein and 10.65g
mobile gum (cold ether extra) and 5.55g stiff gum; in making the 10.65g with 8g pure gum
de got 2.8g more gum 2 + a cone of filtrate adding alcohol ale got a total of 5.55g incl xx full lactate
3 hrs evapor to 336g dl lactate acid; the sq. ale filtrate and gum gum salts only now dissolved
The 16.1g gums taking 50 to the tank 29.06 cc $\therefore$ NaOH (heavy for lactate = 3157cc) $\therefore$ treat rest with 56.0g
Bueinate and got in 3 crops; 21.35g upr 194°; 17.88g upr 185-190°; 2.78g upr 180-185°) a total of 42.01g
1,3 Divy butyric Buein; the ale filtrate united (see above).
1,3 Divy butyric ph. hydrogide from the and combined 13 Divy butyric Buein 42.01g + 76.09g = 118.1g total
these gum with 450g BaM₂ etc 94.6g Buein & 2.12g gum (other extra); this mixed with 22cc ph. hydrogide
and 10cc alcohol gum 42 hrs O.T. seeding ale 17.48g more 1,3 Divy butyric ph. hydrogide, upr 128-132
next remained same on mixing with a prep from L. Anabrusse (May 14-08 = 15.24g) & about 0.5g was
found $\pm$ absol in a one de tube $\rho = 4.0$ 2. = 2.
The 5.55g stiff gum + 3.65g gum A' $\therefore$ total = 9.2g were now exp in open dish for 16 hrs with H₂O ale +
got back 7.15g gum ($\frac{1}{2}$ th tank 23.75cc $\therefore$ NaOH) $\therefore$ treat with H₂O ale upr for 23 hrs longer +
got back 5.7g gum which with 17g Buein exp 3x $\therefore$ 6.16g etc gum 4.24g of dl Buein salt upr
195-210°; the ale filtrate now combined (see above) worked up for free gum
G metaseacharins The 23.97g Buein salts obtained in 4 crops as stated above were now xx from 500 H₂O +
100cc alcohol and got very rapidly 11.33g exp I hours gums upr 200-210° which were obviously
pure dl. with no (5 Buein); this gum with 5.0g BaM₂ etc 8.52g Buein & 2.2g gum
(harder extra) + was resolved into $\frac{3}{4}$ + $\frac{1}{4}$ pt; $\frac{1}{4}$ $\therefore$ 0.5g with 0.5cc ph. hydrogide + 0.5cc ale gum
The ale filtrate from exp I now worked up for free gum meeting with
other ale filtrates as stated above

C5 gums and C6 gums: summary C5 gums: lost other ex. waste (p 108)

(a) 7.35g; (b) 5.79g (c) 2.3g gum; besides 3/4 of 2.2g C5 d. l. anther gum whose Bicin salt mpt $200-210^{\circ}$? has it been treated with 4.7g Bicin?

C6 gums 13.05g gums, sol. in hot acetic ether - which were subsequently resolved into 5.90g gum easily sol. in sw. and cold ether acetate and into 6.75g insoluble therein. and C6 gums (ple extr) cont tar 7.0g which were acylated with 21cc anhydride at 100° gave 10.7g acylated gums of which 5.0g tar remained insoluble in ether. Hydrolysis of ether sol acylated gums by 15.0g Ba(OH)₂ + 15cc H₂O then Bicin gum back 3.1g clean C6 gums

The C5 gums (a) & (c) were all treated with Bicin in theory they were C5 lactones & ag. sol. extr 3x by 6 H₂O took out in all only 2.12g Bicin: OK. Ag. sol. dist off in vacuo & xx from alcohol etc are giving various crops of all Bicin salts which are on hand under hood.

The C6 gums (in 3 parts): of the the 5.9g gum & the 6.75g gums were treated with Bicin extr 3x by 6 H₂O etc but up to date had got no xx of Bicin salts after removing H₂O in vacuo & xx from alcohol - (under hood all summer) - Since Σ - 3.1g on hand

(72) Mixture of d. l. threo C5 Chlorine with eq. amt mixture of l. threo d. l. threo Chlorine

Took 20g d. l. threo C5 Chlorine mpt 174° (May 14, 09) and 20g l. threo d. l. threo C5 Chlorine mpt 165° (May 12-09) + dissolved these in 30cc alcohol: on cooling got a separation of 2.2g xx needles mpt 170° and this xxx were from 20cc alcohol. alcohol gum 1.24g (?). d. l. threo C5 Chlorine mpt 174° with [2] $\Sigma = -105.2$ ($p = 4.0$); the alc. filtr from exp I distilled off in vacuo 60° + dissolve salt (1.1g by diff) in 5cc alcohol now gum rapidly 0.35g xx needles mpt $160-165^{\circ}$ - this indicates that a mixture of all 4 C5 metasaccharins ought to be partially separated through xx of their Chlorine salts.

(73) Formose + CarM₂ Nov 24-08 of No 44 - continuation

Took the 96.0g Formose + CarM₂ gum (Sept 15-08) of the when the phos. had not been removed by oxalic acid as was subseq. done before exp the Formose from ag. sol at reduced pressure 7 hrs 200° got 16g acylated gums of which 50.4g remained insol acylated with 300cc anhydride 7 hrs 200° got 16g acylated gums on 1st by dr 62.7g gums - or after 2nd by dr. with 150g in ether & got from ether sol. acylated gums as follows. B & C were quite larry $\therefore$ had to acylate again & got only 16g tar from B and 4.3g tar from C & got back from Ba(OH)₂ got 58.7g Saccharins distributed as follows. B & C combined 20.5g gums

30.55g A (5x 12 R. 7.
17.65g B (acetic ether extr)
10.50g C (alc extr) (hydr. c. Ba(OH) ₂ and)

Had also follow. ants. of Saccharin gum from other Formose expt. (1044)
 6.8g gum from 180-200 mpt. Buein salts; 8.5g gum A: 16.4g B :: total = 31.7g
 also have 7.35g gum C - or 4.78g C after a 2nd acetylation see Nov 24-08
 the 31.7g Formose gums were now united with the 30.53g gums A (see p. 109) & the gums
 dissolved in wce H₂O & 20x by other gave 19.05g ~~30.53g~~ ^{hydr C Buein} ~~and other sol. gums & these~~
 shaken in cold with 200 cc ether all dissolved except 1.65g gum put with aq. sol.
 The cold ether ext. contained aq. det. :: 17.4g C₃ & gums.
 The aq. sol. gum back 39.25g gums which were now united with the 20.5g gums (p. 109)
 & these then ext. 5X R.F. with 500cc ether each time when got a resolution into
 20.8g soluble and fairly mobile C₃ gums and into
 38.0g insol. gums which, with 4.78g gum C ^{above}, is now all clean sacch gums.
 The 38.0g gums were now resolved by digestion R.F. wce ethyl acetate cool etc into
 16.45g easily soluble gums and into 19.7g insol. gums which with the 4.78g C gum
 above makes a total of 24.15g (aq. det) C₃ gums.
 The 20.8g gum was now ext. 4X R.F. by 400cc ether each time and all but 4.55g gum was
 united with 16.45g lot dissolved. By aq. det. 16.3g gums dissolved in the hot ether:
 1/5th of the 16.3g gum took 25.11 cc ^{to} Nava (theory for C₃ lactone = 24.70 cc) :: this from 16.0g
 was heated with 48.0g Buein typ. by XX from ale 20.4g, crop I, mpt 188-198°; all with 20g Buein
 probably since in XX from 7cc H₂O & 100cc alcohol got 10.45g salts mpt 200-208° 4 sided prisms
 for then 13.45g crop II salts mpt 180-183° then no more.
 The C₃ & C₄ gums, 17.43g, on taking 15.26g (0.261g) gum 24.56 cc ^{to} Nava :: total rest of aq. sol
 with 63.0g Buein typ. ^{by XX} mpt 185-198° crop I; 11.78g, crop II, mpt 185-190° crude 1,3 dioxyl butyric
 Buein & then no more: the combined salts :: 24.56g gum with 10g Buein etc 18.18g Buein and
 4.45g gum (other ext.) in mixing this with 4.5cc ph. hydr & 2.5cc ale got 17.42 lbs 1.45g 1,3 dioxyl
 butyric ph. hydrazide mpt 128-132°; brown solid 0.5g was ± opacely in 4.0% conc & also
 found mpt same on mixing with a ^{single} ^{small} ^{scale} from l. Anathose May 14-08
 The ale. filter from the 2 crops Buein salts had 58.0g salts which with 30.0g Buein etc
 gum 41.88g Buein & 5.83g mobile oil (cold ether ext.) when 0.95g gum (hot ether ext.) in solution
 then 3.5g (ale ext.) :: water lost 2 gums = 4.48g. The 5.83g (cold ether ext.) in solution
 taking 0.2332g took 22.34 cc ^{to} Nava :: heated rest of aq. sol with 4.0g ZnO₃ etc (1.35g mpt 200-208°)
 & got a total of 2.6g Zn dl lactate XX 3 H₂O [comp to 156g dl lactic acid, D.F.D.]
 It is evident that a very large amount of these are present and that only the 4
 C₅ gums Cosmetasaccharins are there. tried to get from the 2k. by then & Buein salts melting
 above 200° by Buein etc when by Chlorine a mod. into d. by then & Chlorine which
 is very diff sol. in alcohol & into l. by then & Chlorine which
 must be easily soluble but did not succeed

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probably because the salt had 1.3 oxy butyric acid as well as d l then & given also
present: must repeat in fall; for same reason an expt with ph. hydr + alc diol + 0.5 gm failed
also because C₄ & C₅ isobaccharides are present (cf 1910-13)

(111)

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